

# Naphthalene, 1-pentadecyl-

<b>Other names:</b>	1-«alpha»-Naphthylpentadecane 1-n-Pentadecylnaphthalene 1-Pentadecylnaphthalene
<b>Inchi:</b>	InChI=1S/C25H38/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-18-23-20-17-21-24-19-15-16-22-25
<b>InchiKey:</b>	CRNKGYFOWLZFDQ-UHFFFAOYSA-N
<b>Formula:</b>	C25H38
<b>SMILES:</b>	CCCCCCCCCCCCCCCc1cccc2cccc12
<b>Mol. weight [g/mol]:</b>	338.57
<b>CAS:</b>	55191-63-4

## Physical Properties

Property code	Value	Unit	Source
gf	369.05	kJ/mol	Joback Method
hf	-143.20	kJ/mol	Joback Method
hfus	51.18	kJ/mol	Joback Method
hvap	75.82	kJ/mol	Joback Method
log10ws	-9.52		Crippen Method
logp	8.473		Crippen Method
mcvol	319.890	ml/mol	McGowan Method
pc	1070.06	kPa	Joback Method
tb	822.04	K	Joback Method
tc	1019.78	K	Joback Method
tf	314.75 ± 0.50	K	NIST Webbook
vc	1.250	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1090.55	J/mol×K	986.82	Joback Method
cpg	1105.86	J/mol×K	1019.78	Joback Method
cpg	1001.10	J/mol×K	822.04	Joback Method
cpg	1020.98	J/mol×K	855.00	Joback Method
cpg	1039.78	J/mol×K	887.95	Joback Method
cpg	1057.59	J/mol×K	920.91	Joback Method

cpg	1074.49	J/molxK	953.87	Joback Method
dvisc	0.0001008	Paxs	822.04	Joback Method
dvisc	0.0001278	Paxs	758.89	Joback Method
dvisc	0.0011599	Paxs	443.15	Joback Method
dvisc	0.0005988	Paxs	506.30	Joback Method
dvisc	0.0003580	Paxs	569.45	Joback Method
dvisc	0.0002371	Paxs	632.60	Joback Method
dvisc	0.0001693	Paxs	695.74	Joback Method
hvapt	96.70	kJ/mol	507.00	NIST Webbook
hvapt	98.10	kJ/mol	499.00	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C55191634&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C55191634&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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