

Benzamide, 2-fluoro-3-trifluoromethyl-N-pentyl-N-dodecyl-

Inchi:	InChI=1S/C25H39F4NO/c1-3-5-7-8-9-10-11-12-13-15-20-30(19-14-6-4-2)24(31)21-17-16
InchiKey:	LOMJLKIUCCLNRK-UHFFFAOYSA-N
Formula:	C25H39F4NO
SMILES:	CCCCCCCCCCCCN(CCCCC)C(=O)c1cccc(C(F)(F)F)c1F
Mol. weight [g/mol]:	445.58

Physical Properties

Property code	Value	Unit	Source
gf	-541.77	kJ/mol	Joback Method
hf	-1183.98	kJ/mol	Joback Method
hfus	63.29	kJ/mol	Joback Method
hvap	79.07	kJ/mol	Joback Method
log10ws	-9.33		Crippen Method
logp	8.398		Crippen Method
mcvol	357.980	ml/mol	McGowan Method
pc	856.47	kPa	Joback Method
rinsol	3202.00		NIST Webbook
tb	868.20	K	Joback Method
tc	1062.93	K	Joback Method
tf	510.15	K	Joback Method
vc	1.413	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1170.73	J/mol×K	868.20	Joback Method
cpg	1189.72	J/mol×K	900.65	Joback Method
cpg	1207.60	J/mol×K	933.11	Joback Method
cpg	1224.45	J/mol×K	965.56	Joback Method
cpg	1240.35	J/mol×K	998.02	Joback Method
cpg	1255.39	J/mol×K	1030.47	Joback Method
cpg	1269.63	J/mol×K	1062.93	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U416700&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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