

Phthalic acid, 2,2,3,3,4,4,4-heptafluorobutyl tridecyl ester

Inchi:	InChI=1S/C25H33F7O4/c1-2-3-4-5-6-7-8-9-10-11-14-17-35-21(33)19-15-12-13-16-20(19
InchiKey:	GWQTTZHABXYZ-UHFFFAOYSA-N
Formula:	C25H33F7O4
SMILES:	CCCCCCCCCCCCOC(=O)c1cccc1C(=O)OCC(F)(F)C(F)(F)C(F)(F)F
Mol. weight [g/mol]:	530.52

Physical Properties

Property code	Value	Unit	Source
gf	-1560.59	kJ/mol	Joback Method
hf	-2222.89	kJ/mol	Joback Method
hfus	59.05	kJ/mol	Joback Method
hvap	82.89	kJ/mol	Joback Method
log10ws	-9.53		Crippen Method
logp	8.144		Crippen Method
mcvol	366.620	ml/mol	McGowan Method
pc	831.46	kPa	Joback Method
rinsol	2607.00		NIST Webbook
tb	940.84	K	Joback Method
tc	1154.86	K	Joback Method
tf	566.16	K	Joback Method
vc	1.468	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1230.15	J/molxK	940.84	Joback Method
cpg	1246.41	J/molxK	976.51	Joback Method
cpg	1261.50	J/molxK	1012.18	Joback Method
cpg	1275.54	J/molxK	1047.85	Joback Method
cpg	1288.64	J/molxK	1083.52	Joback Method
cpg	1300.91	J/molxK	1119.19	Joback Method
cpg	1312.47	J/molxK	1154.86	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U415551&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
m cvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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