

# 2,6,10,14-Tetramethyl-7-(3-methyl-pent-4-enyl)-pentadecene

<b>Other names:</b>	5-Pentadecene, 2,6,10,14-tetramethyl-7-(3-methyl-pent-4-enyl)
<b>Inchi:</b>	InChI=1S/C25H48/c1-9-22(6)16-18-25(24(8)15-11-13-21(4)5)19-17-23(7)14-10-12-20(2)
<b>InchiKey:</b>	YJQZREBAXWNCAX-BUVRLJJBSA-N
<b>Formula:</b>	C <sub>25</sub> H <sub>48</sub>
<b>SMILES:</b>	<chem>C=CC(C)CCC(CCC(C)CCCC(C)C)C(C)=CCCC(C)C</chem>
<b>Mol. weight [g/mol]:</b>	348.65

## Physical Properties

Property code	Value	Unit	Source
gf	306.93	kJ/mol	Joback Method
hf	-352.87	kJ/mol	Joback Method
hfus	40.50	kJ/mol	Joback Method
hvap	68.67	kJ/mol	Joback Method
log10ws	-8.79		Crippen Method
logp	8.830		Crippen Method
mcpol	354.510	ml/mol	McGowan Method
pc	826.21	kPa	Joback Method
rinpol	2072.00		NIST Webbook
rinpol	2074.00		NIST Webbook
rinpol	2079.00		NIST Webbook
rinpol	2079.00		NIST Webbook
rinpol	2072.00		NIST Webbook
tb	769.92	K	Joback Method
tc	950.60	K	Joback Method
tf	275.71	K	Joback Method
vc	1.367	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1102.87	J/mol×K	769.92	Joback Method
cpg	1125.82	J/mol×K	800.03	Joback Method
cpg	1147.65	J/mol×K	830.15	Joback Method
cpg	1168.41	J/mol×K	860.26	Joback Method

cpg	1188.15	J/mol×K	890.37	Joback Method
cpg	1206.95	J/mol×K	920.48	Joback Method
cpg	1224.85	J/mol×K	950.60	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R500759&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R500759&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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