

# Benzene, [3-(2-cyclohexylethyl)-6-cyclopentylhexyl]-

Other names:

1-Cyclohexyl-6-cyclopentyl-3-phenethylhexane

1-Phenyl-3-(2'-cyclohexylethyl)-6-cyclopentylhexane

Inchi: InChI=1S/C25H40/c1-3-10-23(11-4-1)18-20-25(17-9-16-22-14-7-8-15-22)21-19-24-12-5-

InchiKey: DQHMQFJGANEDIX-UHFFFAOYSA-N

Formula: C25H40

SMILES: c1ccc(CCC(CCCC2CCCC2)CCC2CCCCC2)cc1

Mol. weight [g/mol]: 340.59

CAS: 55334-30-0

## Physical Properties

Property code	Value	Unit	Source
gf	330.59	kJ/mol	Joback Method
hf	-213.28	kJ/mol	Joback Method
hfus	36.79	kJ/mol	Joback Method
hvap	73.82	kJ/mol	Joback Method
log10ws	-8.45		Crippen Method
logp	7.956		Crippen Method
mvol	317.630	ml/mol	McGowan Method
pc	1194.00	kPa	Joback Method
tb	832.47	K	Joback Method
tc	1053.92	K	Joback Method
tf	401.21	K	Joback Method
vc	1.196	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1062.94	J/molxK	832.47	Joback Method
cpg	1167.56	J/molxK	1017.01	Joback Method
cpg	1149.68	J/molxK	980.11	Joback Method
cpg	1130.39	J/molxK	943.20	Joback Method
cpg	1109.57	J/molxK	906.29	Joback Method
cpg	1087.13	J/molxK	869.38	Joback Method
cpg	1184.13	J/molxK	1053.92	Joback Method

dvisc	0.0000729	Paxs	832.47	Joback Method
dvisc	0.0000998	Paxs	760.59	Joback Method
dvisc	0.0001460	Paxs	688.72	Joback Method
dvisc	0.0002333	Paxs	616.84	Joback Method
dvisc	0.0004219	Paxs	544.96	Joback Method
dvisc	0.0009134	Paxs	473.09	Joback Method
dvisc	0.0026083	Paxs	401.21	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.88897e+01
Coeff. B	-9.76874e+03
Temperature range (K), min.	525.14
Temperature range (K), max.	719.44

## Sources

The Yaws Handbook of Vapor Pressure:  
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>  
<http://pubs.acs.org/doi/abs/10.1021/ci990307i>

Crippen Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

Joback Method:

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C55334300&Units=SI>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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