

# Estra-1,3,5(10)-trien-17-one, 1-methyl-3,6,7-tris[(trifluoroacetyl)oxy]-, (6«alpha»,7«alpha»)-

Other names: 1-Methyl-6«alpha»,7«alpha»-dihydroxyestrone, TFA  
Inchi: InChI=1S/C25H21F9O7/c1-9-7-10(39-19(36)23(26,27)28)8-12-15(9)11-5-6-22(2)13(3-4-10)/p-1  
InchiKey: IXDHLXAZUNPBDV-HTUBNKTISA-N  
Formula: C25H21F9O7  
SMILES: Cc1cc(OC(=O)C(F)(F)F)cc2c1C1CCC3(C)C(=O)CCC3C1C(OC(=O)C(F)(F)F)C2OC(=O)C(F)(F)F  
Mol. weight [g/mol]: 604.42  
CAS: 56588-13-7

## Physical Properties

Property code	Value	Unit	Source
gf	-2196.55	kJ/mol	Joback Method
hf	-2860.25	kJ/mol	Joback Method
hfus	53.85	kJ/mol	Joback Method
hvap	93.98	kJ/mol	Joback Method
log10ws	-7.38		Crippen Method
logp	5.576		Crippen Method
mcvol	346.590	ml/mol	McGowan Method
pc	1026.63	kPa	Joback Method
rinpol	2520.00		NIST Webbook
tb	1108.44	K	Joback Method
tc	1359.36	K	Joback Method
tf	790.72	K	Joback Method
vc	1.385	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1294.07	J/molxK	1108.44	Joback Method
cpg	1314.36	J/molxK	1150.26	Joback Method
cpg	1335.05	J/molxK	1192.08	Joback Method
cpg	1356.41	J/molxK	1233.90	Joback Method
cpg	1378.75	J/molxK	1275.72	Joback Method
cpg	1402.34	J/molxK	1317.54	Joback Method
cpg	1427.46	J/molxK	1359.36	Joback Method

# Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C56588137&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C56588137&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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