

# Benzene, 1-butyl-2-methyl-

<b>Other names:</b>	1-Butyl-2-methylbenzene 1-Methyl-2-butylbenzene 1-Methyl-2-n-Butylbenzene Toluene, o-butyl-
<b>Inchi:</b>	InChI=1S/C11H16/c1-3-4-8-11-9-6-5-7-10(11)2/h5-7,9H,3-4,8H2,1-2H3
<b>InchiKey:</b>	NUJILYKLNKQOOX-UHFFFAOYSA-N
<b>Formula:</b>	C11H16
<b>SMILES:</b>	CCCCc1ccccc1C
<b>Mol. weight [g/mol]:</b>	148.24
<b>CAS:</b>	1595-11-5

## Physical Properties

Property code	Value	Unit	Source
gf	144.52	kJ/mol	Joback Method
hf	-45.31	kJ/mol	Joback Method
hfus	17.90	kJ/mol	Joback Method
hvap	43.02	kJ/mol	Joback Method
log10ws	-3.59		Crippen Method
logp	3.338		Crippen Method
mcvol	142.090	ml/mol	McGowan Method
pc	2608.40	kPa	Joback Method
rinpol	1154.00		NIST Webbook
rinpol	1161.00		NIST Webbook
ripol	1396.00		NIST Webbook
ripol	1395.80		NIST Webbook
tb	482.74	K	Joback Method
tc	685.93	K	Joback Method
tf	252.67	K	Joback Method
vc	0.543	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	300.77	J/molxK	482.74	Joback Method

cpg	316.42	J/molxK	516.60	Joback Method
cpg	331.27	J/molxK	550.47	Joback Method
cpg	345.33	J/molxK	584.33	Joback Method
cpg	358.64	J/molxK	618.20	Joback Method
cpg	371.23	J/molxK	652.06	Joback Method
cpg	383.12	J/molxK	685.93	Joback Method
dvisc	0.0025675	Paxs	252.67	Joback Method
dvisc	0.0012843	Paxs	291.01	Joback Method
dvisc	0.0007549	Paxs	329.36	Joback Method
dvisc	0.0004957	Paxs	367.71	Joback Method
dvisc	0.0003524	Paxs	406.05	Joback Method
dvisc	0.0002657	Paxs	444.39	Joback Method
dvisc	0.0002096	Paxs	482.74	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.44020e+01
Coeff. B	-3.98664e+03
Coeff. C	-7.36700e+01
Temperature range (K), min.	356.12
Temperature range (K), max.	512.22

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1595115&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1595115&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>ripol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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