

# 13-Docosen-1-ol, (Z)-

Other names:

(Z)-docos-13-enol  
DOCOSEN-(13)-OL-(1)  
cis-13-docosen-1-ol

Inchi:

InChI=1S/C22H44O/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-19-20-21-22-23/h9-

InchiKey:

CFOQKXQWGLAKSK-KTKRTIGZSA-N

Formula:

C22H44O

SMILES:

CCCCCCCCC=CCCCCCCCCCCCCO

Mol. weight [g/mol]:

324.58

CAS:

629-98-1

## Physical Properties

Property code	Value	Unit	Source
gf	77.76	kJ/mol	Joback Method
hf	-532.42	kJ/mol	Joback Method
hfus	57.03	kJ/mol	Joback Method
hvap	81.20	kJ/mol	Joback Method
log10ws	-8.15		Crippen Method
logp	7.577		Crippen Method
mcvol	322.410	ml/mol	McGowan Method
pc	985.16	kPa	Joback Method
rinpol	2466.50		NIST Webbook
rinpol	2466.50		NIST Webbook
tb	799.10	K	Joback Method
tc	978.43	K	Joback Method
tf	393.44	K	Joback Method
vc	1.266	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1025.43	J/mol×K	799.10	Joback Method
cpg	1045.28	J/mol×K	828.99	Joback Method
cpg	1064.20	J/mol×K	858.88	Joback Method
cpg	1082.23	J/mol×K	888.76	Joback Method

cpg	1099.43	J/molxK	918.65	Joback Method
cpg	1115.85	J/molxK	948.54	Joback Method
cpg	1131.53	J/molxK	978.43	Joback Method
dvisc	0.0022519	Paxs	393.44	Joback Method
dvisc	0.0004926	Paxs	461.05	Joback Method
dvisc	0.0001589	Paxs	528.66	Joback Method
dvisc	0.0000663	Paxs	596.27	Joback Method
dvisc	0.0000330	Paxs	663.88	Joback Method
dvisc	0.0000187	Paxs	731.49	Joback Method
dvisc	0.0000117	Paxs	799.10	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.62563e+01
Coeff. B	-6.56216e+03
Coeff. C	-1.37182e+02
Temperature range (K), min.	548.12
Temperature range (K), max.	736.75

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C629981&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C629981&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>r in pol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/72-413-3/13-Docosen-1-ol-Z.pdf>

Generated by Cheméo on 2024-04-23 14:58:39.960308775 +0000 UTC m=+16173568.880886087.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.