

# Benzene, 1-ethyl-2,4,5-trimethyl-

<b>Other names:</b>	1,2,4-Trimethyl,5-Ethylbenzene 1-Ethyl-2,4,5-trimethylbenzene 5-Ethyl-1,2,4-trimethylbenzene Benzene, 1,2,4-trimethyl-5-ethyl Benzene, 5-ethyl-1,2,4-trimethyl
<b>Inchi:</b>	InChI=1S/C11H16/c1-5-11-7-9(3)8(2)6-10(11)4/h6-7H,5H2,1-4H3
<b>InchiKey:</b>	IYFUQKGDILUVJG-UHFFFAOYSA-N
<b>Formula:</b>	C11H16
<b>SMILES:</b>	CCc1cc(C)c(C)cc1C
<b>Mol. weight [g/mol]:</b>	148.24
<b>CAS:</b>	17851-27-3

## Physical Properties

Property code	Value	Unit	Source
gf	125.26	kJ/mol	Joback Method
hf	-68.25	kJ/mol	Joback Method
hfus	17.12	kJ/mol	Joback Method
hvap	44.34	kJ/mol	Joback Method
log10ws	-3.70		Crippen Method
logp	3.174		Crippen Method
mcvol	142.090	ml/mol	McGowan Method
pc	2530.27	kPa	Joback Method
rinpol	1173.00		NIST Webbook
rinpol	1173.00		NIST Webbook
ripol	1391.00		NIST Webbook
ripol	1391.00		NIST Webbook
ripol	1391.00		NIST Webbook
ripol	1390.80		NIST Webbook
tb	484.00 ± 5.00	K	NIST Webbook
tb	484.00 ± 5.00	K	NIST Webbook
tc	698.35	K	Joback Method
tf	260.15 ± 0.50	K	NIST Webbook
tf	259.40 ± 1.50	K	NIST Webbook
tf	259.57 ± 0.20	K	NIST Webbook
vc	0.543	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	381.26	J/molxK	698.35	Joback Method
cpg	302.04	J/molxK	492.70	Joback Method
cpg	316.91	J/molxK	526.98	Joback Method
cpg	331.09	J/molxK	561.25	Joback Method
cpg	344.60	J/molxK	595.53	Joback Method
cpg	357.45	J/molxK	629.80	Joback Method
cpg	369.67	J/molxK	664.08	Joback Method
dvisc	0.0001943	Paxs	492.70	Joback Method
dvisc	0.0013334	Paxs	277.71	Joback Method
dvisc	0.0008052	Paxs	313.54	Joback Method
dvisc	0.0005392	Paxs	349.37	Joback Method
dvisc	0.0003891	Paxs	385.21	Joback Method
dvisc	0.0002967	Paxs	421.04	Joback Method
dvisc	0.0002362	Paxs	456.87	Joback Method
hvapt	56.40	kJ/mol	399.00	NIST Webbook

## Sources

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C17851273&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**KDB:**

<https://www.cheric.org/files/research/kdb/mol/mol695.mol>

## Legend

**cpg:** Ideal gas heat capacity

**dvisc:** Dynamic viscosity

**gf:** Standard Gibbs free energy of formation

**hf:** Enthalpy of formation at standard conditions

**hfus:** Enthalpy of fusion at standard conditions

**hvap:** Enthalpy of vaporization at standard conditions

<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>ripola:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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