

Heptafluorobutanamide, N,N-dinonyl-

Inchi: InChI=1S/C22H38F7NO/c1-3-5-7-9-11-13-15-17-30(18-16-14-12-10-8-6-4-2)19(31)20(23)
InchiKey: GJIHYWPLADFLCD-UHFFFAOYSA-N
Formula: C22H38F7NO
SMILES: CCCCCCCCCN(CCCCCCCC)C(=O)C(F)(F)C(F)(F)C(F)(F)F
Mol. weight [g/mol]: 465.53

Physical Properties

Property code	Value	Unit	Source
gf	-1238.93	kJ/mol	Joback Method
hf	-1941.48	kJ/mol	Joback Method
hfus	56.67	kJ/mol	Joback Method
hvap	63.75	kJ/mol	Joback Method
log10ws	-8.67		Crippen Method
logp	8.149		Crippen Method
mcvol	344.780	ml/mol	McGowan Method
pc	803.88	kPa	Joback Method
rinpola	2098.00		NIST Webbook
rinpola	2098.00		NIST Webbook
tb	754.27	K	Joback Method
tc	923.44	K	Joback Method
tf	431.49	K	Joback Method
vc	1.385	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1097.18	J/molxK	754.27	Joback Method
cpg	1116.36	J/molxK	782.46	Joback Method
cpg	1134.51	J/molxK	810.66	Joback Method
cpg	1151.71	J/molxK	838.85	Joback Method
cpg	1168.02	J/molxK	867.05	Joback Method
cpg	1183.53	J/molxK	895.24	Joback Method
cpg	1198.31	J/molxK	923.44	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U308277&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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