

# Tricyclo[4.3.1.1(2,5)]undec-3-en-10-ol, stereoisomer

Inchi:	InChI=1S/C11H16O/c12-11-9-2-1-3-10(11)8-5-4-7(9)6-8/h4-5,7-12H,1-3,6H2
InchiKey:	XKEDIGAHHBTICC-UHFFFAOYSA-N
Formula:	C11H16O
SMILES:	OC1C2CCCC1C1C=CC2C1
Mol. weight [g/mol]:	164.24
CAS:	70220-95-0

## Physical Properties

Property code	Value	Unit	Source
gf	77.51	kJ/mol	Joback Method
hf	-199.42	kJ/mol	Joback Method
hfus	21.90	kJ/mol	Joback Method
hvap	56.52	kJ/mol	Joback Method
ie	8.40	eV	NIST Webbook
ie	8.82	eV	NIST Webbook
log10ws	-2.37		Crippen Method
logp	1.969		Crippen Method
mcvol	134.840	ml/mol	McGowan Method
pc	3184.73	kPa	Joback Method
tb	561.84	K	Joback Method
tc	767.41	K	Joback Method
tf	313.61	K	Joback Method
vc	0.509	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	372.26	J/molxK	561.84	Joback Method
cpg	389.72	J/molxK	596.10	Joback Method
cpg	406.04	J/molxK	630.36	Joback Method
cpg	421.30	J/molxK	664.62	Joback Method
cpg	435.57	J/molxK	698.89	Joback Method
cpg	448.93	J/molxK	733.15	Joback Method
cpg	461.45	J/molxK	767.41	Joback Method

dvisc	0.0056192	Paxs	313.61	Joback Method
dvisc	0.0033975	Paxs	354.98	Joback Method
dvisc	0.0022818	Paxs	396.35	Joback Method
dvisc	0.0016522	Paxs	437.72	Joback Method
dvisc	0.0012649	Paxs	479.10	Joback Method
dvisc	0.0010104	Paxs	520.47	Joback Method
dvisc	0.0008343	Paxs	561.84	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C70220950&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C70220950&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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