

# 1,8-Paracyclophane

<b>Inchi:</b>	InChI=1S/C21H26/c1-2-4-6-8-19-11-15-21(16-12-19)17-20-13-9-18(10-14-20)7-5-3-1/h9
<b>InchiKey:</b>	DMCPGMNUZQZGNM-UHFFFAOYSA-N
<b>Formula:</b>	C21H26
<b>SMILES:</b>	<chem>c1cc2ccc1CCCCCCCCc1ccc(cc1)C2</chem>
<b>Mol. weight [g/mol]:</b>	278.43
<b>CAS:</b>	6169-94-4

## Physical Properties

Property code	Value	Unit	Source
chs	-11898.00 ± 7.10	kJ/mol	NIST Webbook
gf	327.36	kJ/mol	Joback Method
hf	29.00 ± 9.20	kJ/mol	NIST Webbook
hfs	-82.00 ± 7.10	kJ/mol	NIST Webbook
hfus	21.91	kJ/mol	Joback Method
hsub	111.00	kJ/mol	NIST Webbook
hsub	111.00 ± 2.00	kJ/mol	NIST Webbook
hsub	110.90 ± 2.10	kJ/mol	NIST Webbook
hvap	69.47	kJ/mol	Joback Method
log10ws	-6.75		Crippen Method
logp	5.717		Crippen Method
mvol	248.370	ml/mol	McGowan Method
pc	1908.57	kPa	Joback Method
tb	780.23	K	Joback Method
tc	1048.60	K	Joback Method
tf	405.37	K	Joback Method
vc	0.905	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	857.68	J/mol×K	1048.60	Joback Method
cpg	844.58	J/mol×K	1003.87	Joback Method
cpg	829.57	J/mol×K	959.14	Joback Method
cpg	812.55	J/mol×K	914.42	Joback Method

cpg	793.41	J/mol×K	869.69	Joback Method
cpg	772.05	J/mol×K	824.96	Joback Method
cpg	748.39	J/mol×K	780.23	Joback Method
dvisc	0.0012679	Paxs	405.37	Joback Method
dvisc	0.0000265	Paxs	780.23	Joback Method
dvisc	0.0000381	Paxs	717.75	Joback Method
dvisc	0.0000588	Paxs	655.28	Joback Method
dvisc	0.0000994	Paxs	592.80	Joback Method
dvisc	0.0001901	Paxs	530.32	Joback Method
dvisc	0.0004326	Paxs	467.85	Joback Method
hsubt	105.00 ± 1.30	kJ/mol	365.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.67687e+01
Coeff. B	-7.64741e+03
Temperature range (K), min.	464.01
Temperature range (K), max.	667.47

## Sources

**The Yaws Handbook of Vapor Pressure:**  
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>  
<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C6169944&Units=SI>

## Legend

**chs:** Standard solid enthalpy of combustion  
**cpg:** Ideal gas heat capacity  
**dvisc:** Dynamic viscosity

<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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