

# Propionic acid, 3-iodo-, octadecyl ester

<b>Inchi:</b>	InChI=1S/C21H41IO2/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-20-24-21(23)18-19-2
<b>InchiKey:</b>	XOHCTGSZYYP-POD-UHFFFAOYSA-N
<b>Formula:</b>	C21H41IO2
<b>SMILES:</b>	CCCCCCCCCCCCCCCCCOC(=O)CCI
<b>Mol. weight [g/mol]:</b>	452.45

## Physical Properties

Property code	Value	Unit	Source
gf	-49.86	kJ/mol	Joback Method
hf	-644.70	kJ/mol	Joback Method
hfus	57.34	kJ/mol	Joback Method
hvap	80.87	kJ/mol	Joback Method
log10ws	-8.42		Crippen Method
logp	7.616		Crippen Method
mcvol	340.010	ml/mol	McGowan Method
pc	970.49	kPa	Joback Method
rinsol	2785.00		NIST Webbook
tb	849.31	K	Joback Method
tc	1042.70	K	Joback Method
tf	456.65	K	Joback Method
vc	1.323	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1041.16	J/molxK	849.31	Joback Method
cpg	1124.10	J/molxK	1010.47	Joback Method
cpg	1109.42	J/molxK	978.24	Joback Method
cpg	1093.84	J/molxK	946.00	Joback Method
cpg	1077.30	J/molxK	913.77	Joback Method
cpg	1059.75	J/molxK	881.54	Joback Method
cpg	1137.90	J/molxK	1042.70	Joback Method
dvisc	0.0000453	Paxs	849.31	Joback Method
dvisc	0.0000610	Paxs	783.87	Joback Method

dvisc	0.0000865	Paxs	718.42	Joback Method
dvisc	0.0001316	Paxs	652.98	Joback Method
dvisc	0.0002200	Paxs	587.54	Joback Method
dvisc	0.0004184	Paxs	522.09	Joback Method
dvisc	0.0009562	Paxs	456.65	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U406249&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U406249&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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