

# [2.2.2.2](1,2,3,5)Cyclophane

<b>Inchi:</b>	InChI=1S/C20H20/c1-2-14-11-17-5-3-15-9-13(1)10-16-4-6-18(12-14)20(17)8-7-19(15)16
<b>InchiKey:</b>	WHLLKTALFBGZGG-UHFFFAOYSA-N
<b>Formula:</b>	C20H20
<b>SMILES:</b>	c1c2cc3c4c1CCc1cc(cc(c1CC4)CC3)CC2
<b>Mol. weight [g/mol]:</b>	260.37
<b>CAS:</b>	61477-04-1

## Physical Properties

Property code	Value	Unit	Source
gf	502.04	kJ/mol	Joback Method
hf	233.69	kJ/mol	Joback Method
hfus	26.60	kJ/mol	Joback Method
hvap	69.13	kJ/mol	Joback Method
ie	7.75 ± 0.02	eV	NIST Webbook
log10ws	-5.88		Crippen Method
logp	3.768		Crippen Method
mcvol	212.560	ml/mol	McGowan Method
pc	2263.26	kPa	Joback Method
tb	770.20	K	Joback Method
tc	1024.71	K	Joback Method
tf	513.18	K	Joback Method
vc	0.822	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	625.77	J/molxK	770.20	Joback Method
cpg	643.53	J/molxK	812.62	Joback Method
cpg	660.28	J/molxK	855.04	Joback Method
cpg	676.27	J/molxK	897.45	Joback Method
cpg	691.75	J/molxK	939.87	Joback Method
cpg	706.97	J/molxK	982.29	Joback Method
cpg	722.18	J/molxK	1024.71	Joback Method
dvisc	0.0033005	Paxs	513.18	Joback Method

dvisc	0.0028682	Paxs	556.02	Joback Method
dvisc	0.0025431	Paxs	598.85	Joback Method
dvisc	0.0022914	Paxs	641.69	Joback Method
dvisc	0.0020916	Paxs	684.53	Joback Method
dvisc	0.0019300	Paxs	727.36	Joback Method
dvisc	0.0017968	Paxs	770.20	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C61477041&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C61477041&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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