

# meso-3,4-Dicyclohexyl-2,5-dimethylhexane

**Other names:**

Cyclohexane, 1,1'-(1,2-diethyl-1,2-dimethyl-1,2-ethanediyl)bis-, (R\*,S\*)-(2-Cyclohexyl-1-isopropyl-3-methylbutyl)cyclohexane, meso-Meso-3,4-dicyclohexyl-2,5-dimethylhexan

**Inchi:** InChI=1S/C20H38/c1-15(2)19(17-11-7-5-8-12-17)20(16(3)4)18-13-9-6-10-14-18/h15-20H**InchiKey:** FBRPCXVEEDWFMC-UHFFFAOYSA-N**Formula:** C20H38**SMILES:** CC(C)C(C1CCCCC1)C(C(C)C)C1CCCCC1**Mol. weight [g/mol]:** 278.52**CAS:** 62678-52-8

## Physical Properties

Property code	Value	Unit	Source
chs	-12887.20 ± 1.80	kJ/mol	NIST Webbook
gf	156.66	kJ/mol	Joback Method
hf	-300.00 ± 3.00	kJ/mol	NIST Webbook
hfs	-412.70 ± 1.80	kJ/mol	NIST Webbook
hfus	17.13	kJ/mol	Joback Method
hsub	112.70	kJ/mol	NIST Webbook
hvap	59.42	kJ/mol	Joback Method
log10ws	-6.53		Crippen Method
logp	6.691		Crippen Method
mcvol	270.940	ml/mol	McGowan Method
pc	1375.82	kPa	Joback Method
tb	694.34	K	Joback Method
tc	912.33	K	Joback Method
tf	269.92	K	Joback Method
vc	0.998	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	982.07	J/mol×K	912.33	Joback Method
cpg	962.40	J/mol×K	876.00	Joback Method
cpg	941.16	J/mol×K	839.67	Joback Method

cpg	918.27	J/mol×K	803.33	Joback Method
cpg	893.66	J/mol×K	767.00	Joback Method
cpg	867.26	J/mol×K	730.67	Joback Method
cpg	839.00	J/mol×K	694.34	Joback Method
dvisc	0.0281491	Paxs	269.92	Joback Method
dvisc	0.0000792	Paxs	694.34	Joback Method
dvisc	0.0001210	Paxs	623.60	Joback Method
dvisc	0.0002060	Paxs	552.87	Joback Method
dvisc	0.0004100	Paxs	482.13	Joback Method
dvisc	0.0010339	Paxs	411.39	Joback Method
dvisc	0.0038281	Paxs	340.66	Joback Method
hsubt	113.00 ± 2.10	kJ/mol	343.00	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C62678528&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C62678528&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/72-685-2/meso-3-4-Dicyclohexyl-2-5-dimethylhexane.pdf>

Generated by Cheméo on 2024-04-17 16:04:01.83914159 +0000 UTC m=+15659090.759718901.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.