

# Benzene, (1-methyltridecyl)-

<b>Other names:</b>	Tetradecane, 2-phenyl- 2-Phenyltetradecane
<b>Inchi:</b>	InChI=1S/C20H34/c1-3-4-5-6-7-8-9-10-11-13-16-19(2)20-17-14-12-15-18-20/h12,14-15,1
<b>InchiKey:</b>	GDFUGKICRHMMOT-UHFFFAOYSA-N
<b>Formula:</b>	C20H34
<b>SMILES:</b>	CCCCCCCCCCCC(C)c1ccccc1
<b>Mol. weight [g/mol]:</b>	274.48
<b>CAS:</b>	4534-59-2

## Physical Properties

Property code	Value	Unit	Source
gf	227.49	kJ/mol	Joback Method
hf	-224.88	kJ/mol	Joback Method
hfus	38.07	kJ/mol	Joback Method
hvap	62.00	kJ/mol	Joback Method
log10ws	-7.26		Crippen Method
logp	7.101		Crippen Method
mvol	268.900	ml/mol	McGowan Method
pc	1276.42	kPa	Joback Method
tb	683.24	K	Joback Method
tc	868.95	K	Joback Method
tf	266.00 ± 1.00	K	NIST Webbook
vc	1.042	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	767.49	J/mol×K	683.24	Joback Method
cpg	860.06	J/mol×K	837.99	Joback Method
cpg	843.51	J/mol×K	807.04	Joback Method
cpg	826.02	J/mol×K	776.09	Joback Method
cpg	807.55	J/mol×K	745.14	Joback Method
cpg	788.06	J/mol×K	714.19	Joback Method
cpg	875.71	J/mol×K	868.95	Joback Method

dvisc	0.0000905	Paxs	683.24	Joback Method
dvisc	0.0001241	Paxs	623.80	Joback Method
dvisc	0.0001819	Paxs	564.35	Joback Method
dvisc	0.0002918	Paxs	504.91	Joback Method
dvisc	0.0005310	Paxs	445.47	Joback Method
dvisc	0.0011617	Paxs	386.02	Joback Method
dvisc	0.0033798	Paxs	326.58	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4534592&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4534592&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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