

# N,N-Ethylene bis(N',N,-diphenyl urea)

<b>Inchi:</b>	InChI=1S/C28H26N4O2/c33-27(31(23-13-5-1-6-14-23)24-15-7-2-8-16-24)29-21-22-30-28
<b>InchiKey:</b>	JMJOGHWYXVODDY-UHFFFAOYSA-N
<b>Formula:</b>	C28H26N4O2
<b>SMILES:</b>	O=C(NCCNC(=O)N(c1ccccc1)c1ccccc1)N(c1ccccc1)c1ccccc1
<b>Mol. weight [g/mol]:</b>	450.53
<b>CAS:</b>	6123-88-2

## Physical Properties

Property code	Value	Unit	Source
chs	-14539.00 ± 15.00	kJ/mol	NIST Webbook
gf	777.02	kJ/mol	Joback Method
hf	341.71	kJ/mol	Joback Method
hfs	-195.00 ± 15.00	kJ/mol	NIST Webbook
hfus	63.88	kJ/mol	Joback Method
hvap	117.48	kJ/mol	Joback Method
log10ws	-7.25		Crippen Method
logp	6.082		Crippen Method
mcvol	353.400	ml/mol	McGowan Method
pc	1687.95	kPa	Joback Method
tb	1179.72	K	Joback Method
tc	1449.62	K	Joback Method
tf	781.12	K	Joback Method
vc	1.290	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1175.66	J/molxK	1179.72	Joback Method
cpg	1188.14	J/molxK	1224.70	Joback Method
cpg	1200.36	J/molxK	1269.69	Joback Method
cpg	1212.59	J/molxK	1314.67	Joback Method
cpg	1225.10	J/molxK	1359.65	Joback Method
cpg	1238.16	J/molxK	1404.64	Joback Method
cpg	1252.05	J/molxK	1449.62	Joback Method

# Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C6123882&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C6123882&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

# Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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