

# 2,4-Bis(dimethylbenzyl)-6-t-butylphenol

<b>Other names:</b>	2-(Tert-butyl)-4,6-bis(2-phenylpropan-2-yl)phenol
<b>Inchi:</b>	InChI=1S/C28H34O/c1-26(2,3)23-18-22(27(4,5)20-14-10-8-11-15-20)19-24(25(23)29)28
<b>InchiKey:</b>	QIJHCZVVJYQNJA-UHFFFAOYSA-N
<b>Formula:</b>	C28H34O
<b>SMILES:</b>	CC(C)(C)c1cc(C(C)(C)c2ccccc2)cc(C(C)(C)c2ccccc2)c1O
<b>Mol. weight [g/mol]:</b>	386.57
<b>CAS:</b>	244080-16-8

## Physical Properties

Property code	Value	Unit	Source
gf	356.75	kJ/mol	Joback Method
hf	-138.16	kJ/mol	Joback Method
hfus	33.16	kJ/mol	Joback Method
hvap	95.20	kJ/mol	Joback Method
log10ws	-7.56		Crippen Method
logp	7.341		Crippen Method
mcvol	339.970	ml/mol	McGowan Method
pc	1334.91	kPa	Joback Method
rinpol	2547.30		NIST Webbook
rinpol	2547.30		NIST Webbook
tb	1000.97	K	Joback Method
tc	1262.85	K	Joback Method
tf	628.60	K	Joback Method
vc	1.212	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1129.50	J/molxK	1000.97	Joback Method
cpg	1149.86	J/molxK	1044.62	Joback Method
cpg	1170.05	J/molxK	1088.26	Joback Method
cpg	1190.42	J/molxK	1131.91	Joback Method
cpg	1211.31	J/molxK	1175.56	Joback Method
cpg	1233.08	J/molxK	1219.21	Joback Method

cpg	1256.06	J/mol×K	1262.85	Joback Method
dvisc	0.0000205	Paxs	628.60	Joback Method
dvisc	0.0000085	Paxs	690.66	Joback Method
dvisc	0.0000041	Paxs	752.72	Joback Method
dvisc	0.0000022	Paxs	814.79	Joback Method
dvisc	0.0000013	Paxs	876.85	Joback Method
dvisc	0.0000008	Paxs	938.91	Joback Method
dvisc	0.0000005	Paxs	1000.97	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C244080168&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C244080168&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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