

Copper acetate, hydrate

Other names:	copper acetate monohydrate copper(II) acetate hydrate
Inchi:	InChI=1S/2C2H4O2.Cu.H2O/c2*1-2(3)4;/h2*1H3,(H,3,4);;1H2/q;,+2;/p-2
InchiKey:	NWFNSTOSIVLCJA-UHFFFAOYSA-L
Formula:	C4H8CuO5
SMILES:	CC(=O)[O-].CC(=O)[O-].O.[Cu+2]
Mol. weight [g/mol]:	199.65

Sources

NIST Webbook: <http://webbook.nist.gov/cgi/cbook.cgi?ID=B6005730&Units=SI>

Low-temperature heat capacities and standard molar enthalpy of formation <https://www.doi.org/10.1016/j.tca.2008.02.024>

thermochemistry of coordination <https://www.doi.org/10.1016/j.tca.2011.02.008>

K₂(M(DPA)₂·7H₂O) (M = Cu and Ni, <https://www.doi.org/10.1016/j.tca.2012.12.024>

synthesis, Pyridine 2,6-dicarboxylate

Pyridine 2,6-dicarboxylate:

Latest version available from:

<https://www.cheméo.com/cid/72-893-1/Copper-acetate-hydrate.pdf>

Generated by Cheméo on 2024-04-27 14:08:04.764528938 +0000 UTC m=+16516133.685106254.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.