

# Difluoro-bis(2-fluoro-2,2-dinitroethoxy)methane

Inchi:	lnChI=1S/C5H4F4N4O10/c6-3(10(14)15,11(16)17)1-22-5(8,9)23-2-4(7,12(18)19)13(20)2
InchiKey:	ZRXYSSKJEJXCDB-UHFFFAOYSA-N
Formula:	C5H4F4N4O10
SMILES:	O=[N+]([O-])C(F)(CO(F)(OC(F)([N+]([O-])=O)[N+]([O-])=O)[N+]([O-])=O)[O-]
Mol. weight [g/mol]:	356.10
CAS:	58715-08-5

## Physical Properties

Property code	Value	Unit	Source
chl	-1982.90 ± 6.70	kJ/mol	NIST Webbook
gf	-847.30	kJ/mol	Joback Method
hf	-1264.70	kJ/mol	Joback Method
hfl	-1297.90 ± 6.70	kJ/mol	NIST Webbook
hfus	46.60	kJ/mol	Joback Method
hvap	90.75	kJ/mol	Joback Method
log10ws	-3.64		Crippen Method
logp	-0.077		Crippen Method
mcvol	169.810	ml/mol	McGowan Method
pc	3062.55	kPa	Joback Method
tb	953.39	K	Joback Method
tc	1210.98	K	Joback Method
tf	774.63	K	Joback Method
vc	0.719	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	526.83	J/mol×K	953.39	Joback Method
cpg	531.52	J/mol×K	996.32	Joback Method
cpg	535.63	J/mol×K	1039.25	Joback Method
cpg	539.29	J/mol×K	1082.18	Joback Method
cpg	542.62	J/mol×K	1125.11	Joback Method
cpg	545.74	J/mol×K	1168.04	Joback Method
cpg	548.77	J/mol×K	1210.98	Joback Method

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hvapt

72.70

kJ/mol

340.00

NIST Webbook

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## Sources

**Crippen Method:** [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C58715085&Units=SI>

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci990307l>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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