

# Perfluoro-2-methyl-2-pentene

<b>Other names:</b>	Perfluoro-(2-methylpent-2-ene) 2-Pentene, 1,1,1,3,4,4,5,5,5-nonafluoro-2-(trifluoromethyl)- 2-Methylpent-2-ene, perfluoro- 1,1,1,3,4,4,5,5,5-nonafluoro-2-(trifluoromethyl)pent-4-ene
<b>Inchi:</b>	InChI=1S/C6F12/c7-2(3(8,9)6(16,17)18)1(4(10,11)12)5(13,14)15
<b>InchiKey:</b>	FAEGGADNHFKDQX-UHFFFAOYSA-N
<b>Formula:</b>	C6F12
<b>SMILES:</b>	FC(=C(C(F)(F)F)C(F)(F)F)C(F)(F)C(F)(F)F
<b>Mol. weight [g/mol]:</b>	300.05
<b>CAS:</b>	1584-03-8

## Physical Properties

Property code	Value	Unit	Source
chl	-1993.70 ± 4.00	kJ/mol	NIST Webbook
gf	-2263.60	kJ/mol	Joback Method
hf	-2499.00 ± 4.10	kJ/mol	NIST Webbook
hfl	-2522.40 ± 4.10	kJ/mol	NIST Webbook
hfus	16.18	kJ/mol	Joback Method
hvap	23.40 ± 0.40	kJ/mol	NIST Webbook
log10ws	-4.84		Crippen Method
logp	4.532		Crippen Method
mcvol	112.340	ml/mol	McGowan Method
pc	2073.65	kPa	Joback Method
tb	318.92	K	Joback Method
tc	438.47	K	Joback Method
tf	141.14	K	Joback Method
vc	0.525	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	239.46	J/mol×K	318.92	Joback Method
cpg	250.44	J/mol×K	338.85	Joback Method
cpg	260.78	J/mol×K	358.77	Joback Method

cpg	270.51	J/mol×K	378.70	Joback Method
cpg	279.64	J/mol×K	398.62	Joback Method
cpg	288.21	J/mol×K	418.55	Joback Method
cpg	296.23	J/mol×K	438.47	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1584038&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1584038&amp;Units=SI</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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