

4-Penten-1-ol, 3-methyl-

Other names:	3-Methyl-4-penten-1-ol
Inchi:	InChI=1S/C6H12O/c1-3-6(2)4-5-7/h3,6-7H,1,4-5H2,2H3
InchiKey:	VTCQTYOGWYLVES-UHFFFAOYSA-N
Formula:	C6H12O
SMILES:	C=CC(C)CCO
Mol. weight [g/mol]:	100.16
CAS:	51174-44-8

Physical Properties

Property code	Value	Unit	Source
gf	-51.78	kJ/mol	Joback Method
hf	-199.25	kJ/mol	Joback Method
hfus	10.58	kJ/mol	Joback Method
hvap	44.57	kJ/mol	Joback Method
log10ws	-1.21		Crippen Method
logp	1.191		Crippen Method
mcvol	96.970	ml/mol	McGowan Method
pc	3664.21	kPa	Joback Method
tb	425.10	K	Joback Method
tc	592.88	K	Joback Method
tf	201.44	K	Joback Method
vc	0.365	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	192.23	J/mol×K	425.10	Joback Method
cpg	201.41	J/mol×K	453.06	Joback Method
cpg	210.23	J/mol×K	481.03	Joback Method
cpg	218.68	J/mol×K	508.99	Joback Method
cpg	226.77	J/mol×K	536.95	Joback Method
cpg	234.52	J/mol×K	564.91	Joback Method
cpg	241.94	J/mol×K	592.88	Joback Method
dvisc	0.1985438	Paxs	201.44	Joback Method

dvisc	0.0271287	Paxs	238.72	Joback Method
dvisc	0.0063461	Paxs	275.99	Joback Method
dvisc	0.0020977	Paxs	313.27	Joback Method
dvisc	0.0008774	Paxs	350.55	Joback Method
dvisc	0.0004340	Paxs	387.82	Joback Method
dvisc	0.0002428	Paxs	425.10	Joback Method

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.50131e+01
Coeff. B	-3.83367e+03
Coeff. C	-5.69790e+01
Temperature range (K), min.	317.32
Temperature range (K), max.	452.13

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C51174448&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions

log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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