

# Cyclopentanamine, 1-methyl-

<b>Other names:</b>	1-methylcyclopentanamine
<b>Inchi:</b>	InChI=1S/C6H13N/c1-6(7)4-2-3-5-6/h2-5,7H2,1H3
<b>InchiKey:</b>	TWASBYPJZBHZQJ-UHFFFAOYSA-N
<b>Formula:</b>	C6H13N
<b>SMILES:</b>	CC1(N)CCCC1
<b>Mol. weight [g/mol]:</b>	99.17
<b>CAS:</b>	40571-45-7

## Physical Properties

Property code	Value	Unit	Source
gf	97.15	kJ/mol	Joback Method
hf	-57.66	kJ/mol	Joback Method
hfus	4.13	kJ/mol	Joback Method
hvap	38.70	kJ/mol	Joback Method
log10ws	-1.77		Crippen Method
logp	1.278		Crippen Method
mcvol	94.520	ml/mol	McGowan Method
pc	4374.18	kPa	Joback Method
rinpol	789.00		NIST Webbook
rinpol	789.00		NIST Webbook
tb	273.15 ± 4.00	K	NIST Webbook
tc	646.58	K	Joback Method
tf	275.44	K	Joback Method
vc	0.340	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	193.21	J/mol×K	424.73	Joback Method
cpg	208.32	J/mol×K	461.71	Joback Method
cpg	222.22	J/mol×K	498.68	Joback Method
cpg	235.05	J/mol×K	535.66	Joback Method
cpg	246.92	J/mol×K	572.63	Joback Method
cpg	257.96	J/mol×K	609.61	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.37115e+01
Coeff. B	-3.25660e+03
Coeff. C	-5.07240e+01
Temperature range (K), min.	293.32
Temperature range (K), max.	438.41

## Sources

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C40571457&Units=SI>

**The Yaws Handbook of Vapor Pressure:**

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307I>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpolar:</b>	Non-polar retention indices

**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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