

# Pentane, 2-iodo-4-methyl

Inchi:	InChI=1S/C6H13I/c1-5(2)4-6(3)7/h5-6H,4H2,1-3H3
InchiKey:	HUVYLVIBZCVAKG-UHFFFAOYSA-N
Formula:	C6H13I
SMILES:	CC(C)CC(C)I
Mol. weight [g/mol]:	212.07

## Physical Properties

Property code	Value	Unit	Source
gf	52.88	kJ/mol	Joback Method
hf	-100.86	kJ/mol	Joback Method
hfus	8.66	kJ/mol	Joback Method
hvap	37.55	kJ/mol	Joback Method
log10ws	-3.15		Crippen Method
logp	2.856		Crippen Method
mcvol	121.220	ml/mol	McGowan Method
pc	3086.42	kPa	Joback Method
rinpol	1026.00		NIST Webbook
rinpol	1026.00		NIST Webbook
ripol	1171.00		NIST Webbook
ripol	1171.00		NIST Webbook
tb	428.94	K	Joback Method
tc	639.15	K	Joback Method
tf	185.44	K	Joback Method
vc	0.448	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	209.69	J/molxK	428.94	Joback Method
cpg	262.50	J/molxK	604.11	Joback Method
cpg	253.07	J/molxK	569.08	Joback Method
cpg	243.11	J/molxK	534.04	Joback Method
cpg	232.57	J/molxK	499.01	Joback Method
cpg	221.44	J/molxK	463.97	Joback Method

cpg	271.40	J/molxK	639.15	Joback Method
dvisc	0.0003337	Paxs	428.94	Joback Method
dvisc	0.0004588	Paxs	388.36	Joback Method
dvisc	0.0006793	Paxs	347.77	Joback Method
dvisc	0.0011159	Paxs	307.19	Joback Method
dvisc	0.0021319	Paxs	266.61	Joback Method
dvisc	0.0051392	Paxs	226.02	Joback Method
dvisc	0.0182084	Paxs	185.44	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R25576&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R25576&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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