

# 4-Methyl-2-pentenal

<b>Inchi:</b>	InChI=1S/C6H10O/c1-6(2)4-3-5-7/h3-6H,1-2H3/b4-3+
<b>InchiKey:</b>	RIWPMNBTULNXOH-ONEGZZNKSA-N
<b>Formula:</b>	C6H10O
<b>SMILES:</b>	CC(C)C=CC=O
<b>Mol. weight [g/mol]:</b>	98.14
<b>CAS:</b>	5362-56-1

## Physical Properties

Property code	Value	Unit	Source
gf	-22.10	kJ/mol	Joback Method
hf	-140.81	kJ/mol	Joback Method
hfus	10.26	kJ/mol	Joback Method
hvap	35.24	kJ/mol	Joback Method
log10ws	-1.23		Crippen Method
logp	1.397		Crippen Method
mcvol	92.670	ml/mol	McGowan Method
pc	3628.97	kPa	Joback Method
ripol	1095.00		NIST Webbook
ripol	1095.00		NIST Webbook
tb	389.06	K	Joback Method
tc	574.87	K	Joback Method
tf	179.30	K	Joback Method
vc	0.362	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	164.65	J/mol×K	389.06	Joback Method
cpg	209.01	J/mol×K	543.90	Joback Method
cpg	201.02	J/mol×K	512.93	Joback Method
cpg	192.61	J/mol×K	481.96	Joback Method
cpg	183.76	J/mol×K	451.00	Joback Method
cpg	174.45	J/mol×K	420.03	Joback Method
cpg	216.59	J/mol×K	574.87	Joback Method

dvisc	0.0002530	Paxs	389.06	Joback Method
dvisc	0.0003329	Paxs	354.10	Joback Method
dvisc	0.0004653	Paxs	319.14	Joback Method
dvisc	0.0007061	Paxs	284.18	Joback Method
dvisc	0.0012046	Paxs	249.22	Joback Method
dvisc	0.0024463	Paxs	214.26	Joback Method
dvisc	0.0065488	Paxs	179.30	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.43708e+01
Coeff. B	-3.45391e+03
Coeff. C	-5.26700e+01
Temperature range (K), min.	297.92
Temperature range (K), max.	433.92

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R593449&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R593449&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log10 of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>p<sub>c</sub>:</b>	Critical Pressure
<b>p<sub>vap</sub>:</b>	Vapor pressure
<b>ri<sub>pol</sub>:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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