

3-Hexyn-2-ol

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| Other names: | hex-3-yn-2-ol |
| Inchi: | InChI=1S/C6H10O/c1-3-4-5-6(2)7/h6-7H,3H2,1-2H3 |
| InchiKey: | IFCAMPNPKBSTF-UHFFFAOYSA-N |
| Formula: | C6H10O |
| SMILES: | CCC#CC(C)O |
| Mol. weight [g/mol]: | 98.14 |
| CAS: | 109-50-2 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | 63.18 | kJ/mol | Joback Method |
| hf | -52.38 | kJ/mol | Joback Method |
| hfus | 14.98 | kJ/mol | Joback Method |
| hvap | 47.39 | kJ/mol | Joback Method |
| log10ws | -1.50 | | Crippen Method |
| logp | 0.781 | | Crippen Method |
| mcvol | 92.670 | ml/mol | McGowan Method |
| pc | 4249.61 | kPa | Joback Method |
| rinpol | 796.00 | | NIST Webbook |
| rinpol | 797.00 | | NIST Webbook |
| rinpol | 796.00 | | NIST Webbook |
| tb | 437.42 | K | Joback Method |
| tc | 624.36 | K | Joback Method |
| tf | 309.30 | K | Joback Method |
| vc | 0.346 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|--------|---------|-----------------|---------------|
| cpg | 178.90 | J/mol×K | 437.42 | Joback Method |
| cpg | 187.39 | J/mol×K | 468.58 | Joback Method |
| cpg | 195.55 | J/mol×K | 499.73 | Joback Method |
| cpg | 203.37 | J/mol×K | 530.89 | Joback Method |
| cpg | 210.87 | J/mol×K | 562.04 | Joback Method |

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|-----|--------|---------|--------|---------------|
| cpg | 218.05 | J/mol×K | 593.20 | Joback Method |
| cpg | 224.92 | J/mol×K | 624.36 | Joback Method |

Pressure Dependent Properties

| Property code | Value | Unit | Pressure [kPa] | Source |
|---------------|--------|------|----------------|--------------|
| tbrp | 353.00 | K | 8.00 | NIST Webbook |

Correlations

| Information | Value |
|-----------------------------|-------------------------------|
| Property code | pvap |
| Equation | $\ln(P_{vp}) = A + B/(T + C)$ |
| Coeff. A | 1.64464e+01 |
| Coeff. B | -4.34111e+03 |
| Coeff. C | -5.98080e+01 |
| Temperature range (K), min. | 328.15 |
| Temperature range (K), max. | 450.15 |

Sources

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|---|---|
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C109502&Units=SI |
| The Yaws Handbook of Vapor Pressure: | https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure |

Legend

| | |
|-------------|---|
| cpg: | Ideal gas heat capacity |
| gf: | Standard Gibbs free energy of formation |

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|-----------------|---|
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| pvap: | Vapor pressure |
| rinpola: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tbrp: | Boiling point at reduced pressure |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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