

# Propane, 1,1,1,2,3,3,3-heptachloro-

<b>Inchi:</b>	InChI=1S/C3HCl7/c4-1(2(5,6)7)3(8,9)10/h1H
<b>InchiKey:</b>	LZOUTMTXULWVSU-UHFFFAOYSA-N
<b>Formula:</b>	C3HCl7
<b>SMILES:</b>	CIC(C(CI)(CI)CI)C(CI)(CI)CI
<b>Mol. weight [g/mol]:</b>	285.21
<b>CAS:</b>	3849-33-0

## Physical Properties

Property code	Value	Unit	Source
gf	-105.89	kJ/mol	Joback Method
hf	-238.21	kJ/mol	Joback Method
hfus	14.55	kJ/mol	Joback Method
hvap	49.99	kJ/mol	Joback Method
log10ws	-4.47		Crippen Method
logp	4.334		Crippen Method
mcvol	138.810	ml/mol	McGowan Method
pc	3384.14	kPa	Joback Method
rinpol	1380.00		NIST Webbook
rinpol	1380.00		NIST Webbook
tb	523.15	K	Joback Method
tc	777.32	K	Joback Method
tf	322.85	K	Joback Method
vc	0.518	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	211.37	J/mol×K	523.15	Joback Method
cpg	228.13	J/mol×K	734.96	Joback Method
cpg	226.06	J/mol×K	692.60	Joback Method
cpg	223.45	J/mol×K	650.24	Joback Method
cpg	220.21	J/mol×K	607.87	Joback Method
cpg	216.21	J/mol×K	565.51	Joback Method
cpg	229.76	J/mol×K	777.32	Joback Method

dvisc	0.0003261	Paxs	523.15	Joback Method
dvisc	0.0004407	Paxs	489.77	Joback Method
dvisc	0.0006224	Paxs	456.38	Joback Method
dvisc	0.0009282	Paxs	423.00	Joback Method
dvisc	0.0014823	Paxs	389.62	Joback Method
dvisc	0.0025844	Paxs	356.23	Joback Method
dvisc	0.0050546	Paxs	322.85	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3849330&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3849330&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/73-188-3/Propane-1-1-1-2-3-3-3-heptachloro.pdf>

Generated by Cheméo on 2024-04-29 02:13:14.708480942 +0000 UTC m=+16646043.629058258.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.