

7-Octen-2-ol

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|-----------------------------|--|
| Inchi: | InChI=1S/C8H16O/c1-3-4-5-6-7-8(2)9/h3,8-9H,1,4-7H2,2H3 |
| InchiKey: | HSHUHVOEMTVRS-UHFFFAOYSA-N |
| Formula: | C8H16O |
| SMILES: | C=CCCCC(C)O |
| Mol. weight [g/mol]: | 128.21 |
| CAS: | 39546-75-3 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | -34.94 | kJ/mol | Joback Method |
| hf | -240.53 | kJ/mol | Joback Method |
| hfus | 15.76 | kJ/mol | Joback Method |
| hvap | 49.02 | kJ/mol | Joback Method |
| log10ws | -2.40 | | Crippen Method |
| logp | 2.114 | | Crippen Method |
| mcvol | 125.150 | ml/mol | McGowan Method |
| pc | 2940.89 | kPa | Joback Method |
| rinpol | 1064.00 | | NIST Webbook |
| rinpol | 1064.00 | | NIST Webbook |
| tb | 470.86 | K | Joback Method |
| tc | 636.76 | K | Joback Method |
| tf | 223.98 | K | Joback Method |
| vc | 0.477 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|--------|---------|-----------------|---------------|
| cpg | 273.08 | J/mol×K | 470.86 | Joback Method |
| cpg | 284.36 | J/mol×K | 498.51 | Joback Method |
| cpg | 295.19 | J/mol×K | 526.16 | Joback Method |
| cpg | 305.56 | J/mol×K | 553.81 | Joback Method |
| cpg | 315.50 | J/mol×K | 581.46 | Joback Method |
| cpg | 325.02 | J/mol×K | 609.11 | Joback Method |
| cpg | 334.14 | J/mol×K | 636.76 | Joback Method |

| | | | | |
|-------|-----------|------|--------|---------------|
| dvisc | 0.1097275 | Paxs | 223.98 | Joback Method |
| dvisc | 0.0162547 | Paxs | 265.13 | Joback Method |
| dvisc | 0.0040223 | Paxs | 306.27 | Joback Method |
| dvisc | 0.0013856 | Paxs | 347.42 | Joback Method |
| dvisc | 0.0005982 | Paxs | 388.57 | Joback Method |
| dvisc | 0.0003033 | Paxs | 429.71 | Joback Method |
| dvisc | 0.0001732 | Paxs | 470.86 | Joback Method |

Sources

| | |
|------------------------|---|
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C39546753&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.cheméo.com/doc/models/crippen_log10ws |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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