

# Cyclopentene, 3-ethylidene-1-methyl-

<b>Other names:</b>	3-ethylidene-1-methylcyclopentene
<b>Inchi:</b>	InChI=1S/C8H12/c1-3-8-5-4-7(2)6-8/h3,6H,4-5H2,1-2H3/b8-3+
<b>InchiKey:</b>	KNMRUIAIEAZNRY-FPYGCLRLSA-N
<b>Formula:</b>	C8H12
<b>SMILES:</b>	CC=C1C=C(C)CC1
<b>Mol. weight [g/mol]:</b>	108.18
<b>CAS:</b>	62338-00-5

## Physical Properties

Property code	Value	Unit	Source
gf	126.53	kJ/mol	Joback Method
hf	-5.29	kJ/mol	Joback Method
hfus	10.50	kJ/mol	Joback Method
hvap	35.71	kJ/mol	Joback Method
log10ws	-2.77		Crippen Method
logp	2.673		Crippen Method
mcvol	104.120	ml/mol	McGowan Method
pc	3368.44	kPa	Joback Method
ripol	1195.00		NIST Webbook
ripol	1207.00		NIST Webbook
ripol	1163.00		NIST Webbook
ripol	1163.00		NIST Webbook
ripol	1168.00		NIST Webbook
ripol	1178.00		NIST Webbook
tb	413.17	K	Joback Method
tc	617.70	K	Joback Method
tf	218.70	K	Joback Method
vc	0.395	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	190.45	J/mol×K	413.17	Joback Method
cpg	249.57	J/mol×K	583.61	Joback Method

cpg	239.05	J/molxK	549.52	Joback Method
cpg	227.90	J/molxK	515.43	Joback Method
cpg	216.11	J/molxK	481.35	Joback Method
cpg	203.63	J/molxK	447.26	Joback Method
cpg	259.52	J/molxK	617.70	Joback Method
dvisc	0.0002392	Paxs	413.17	Joback Method
dvisc	0.0002917	Paxs	380.76	Joback Method
dvisc	0.0003693	Paxs	348.35	Joback Method
dvisc	0.0004905	Paxs	315.93	Joback Method
dvisc	0.0006953	Paxs	283.52	Joback Method
dvisc	0.0010785	Paxs	251.11	Joback Method
dvisc	0.0019054	Paxs	218.70	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C62338005&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C62338005&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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