

# 1-ethoxy-2,4-hexadiene

<b>Other names:</b>	2,4-Hexadiene, 1-ethoxy
<b>Inchi:</b>	InChI=1S/C8H14O/c1-3-5-6-7-8-9-4-2/h3,5-7H,4,8H2,1-2H3/b5-3+,7-6-
<b>InchiKey:</b>	UNHPVHFNZXRVNU-WZWXSLMZSA-N
<b>Formula:</b>	C8H14O
<b>SMILES:</b>	CC=CC=CCOCC
<b>Mol. weight [g/mol]:</b>	126.20

## Physical Properties

Property code	Value	Unit	Source
gf	71.92	kJ/mol	Joback Method
hf	-106.23	kJ/mol	Joback Method
hfus	18.07	kJ/mol	Joback Method
hvap	35.73	kJ/mol	Joback Method
log10ws	-1.97		Crippen Method
logp	2.155		Crippen Method
mcvol	120.850	ml/mol	McGowan Method
pc	2773.00	kPa	Joback Method
rinpol	940.00		NIST Webbook
tb	413.18	K	Joback Method
tc	593.96	K	Joback Method
tf	191.99	K	Joback Method
vc	0.462	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	227.62	J/molxK	413.18	Joback Method
cpg	239.92	J/molxK	443.31	Joback Method
cpg	251.66	J/molxK	473.44	Joback Method
cpg	262.87	J/molxK	503.57	Joback Method
cpg	273.58	J/molxK	533.70	Joback Method
cpg	283.79	J/molxK	563.83	Joback Method
cpg	293.54	J/molxK	593.96	Joback Method
dvisc	0.0036057	Paxs	191.99	Joback Method

dvisc	0.0013874	Paxs	228.86	Joback Method
dvisc	0.0006959	Paxs	265.72	Joback Method
dvisc	0.0004129	Paxs	302.59	Joback Method
dvisc	0.0002744	Paxs	339.45	Joback Method
dvisc	0.0001976	Paxs	376.32	Joback Method
dvisc	0.0001509	Paxs	413.18	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U365965&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U365965&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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