

# Pentalene, octahydro-1-methyl-

<b>Other names:</b>	2-methylbicyclo[3.3.0]octane
<b>Inchi:</b>	InChI=1S/C9H16/c1-7-5-6-8-3-2-4-9(7)8/h7-9H,2-6H2,1H3
<b>InchiKey:</b>	XWDLMQDCTIPQFS-UHFFFAOYSA-N
<b>Formula:</b>	C9H16
<b>SMILES:</b>	CC1CCC2CCCC12
<b>Mol. weight [g/mol]:</b>	124.22
<b>CAS:</b>	32273-77-1

## Physical Properties

Property code	Value	Unit	Source
gf	114.49	kJ/mol	Joback Method
hf	-116.15	kJ/mol	Joback Method
hfus	12.21	kJ/mol	Joback Method
hvap	35.49	kJ/mol	Joback Method
log10ws	-2.65		Crippen Method
logp	2.833		Crippen Method
mcvol	115.950	ml/mol	McGowan Method
pc	3069.34	kPa	Joback Method
ripol	1031.00		NIST Webbook
tb	422.67	K	Joback Method
tc	629.88	K	Joback Method
tf	215.79	K	Joback Method
vc	0.436	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	241.34	J/molxK	422.67	Joback Method
cpg	260.95	J/molxK	457.20	Joback Method
cpg	279.43	J/molxK	491.74	Joback Method
cpg	296.83	J/molxK	526.27	Joback Method
cpg	313.20	J/molxK	560.81	Joback Method
cpg	328.59	J/molxK	595.34	Joback Method
cpg	343.05	J/molxK	629.88	Joback Method

dvisc	0.0009865	Paxs	215.79	Joback Method
dvisc	0.0008118	Paxs	250.27	Joback Method
dvisc	0.0007004	Paxs	284.75	Joback Method
dvisc	0.0006238	Paxs	319.23	Joback Method
dvisc	0.0005683	Paxs	353.71	Joback Method
dvisc	0.0005263	Paxs	388.19	Joback Method
dvisc	0.0004936	Paxs	422.67	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C32273771&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C32273771&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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