

# Bicyclo[2.2.1]heptane, 2-ethenyl-

<b>Other names:</b>	2-Vinylbicyclo[2.2.1]heptane
<b>Inchi:</b>	InChI=1S/C9H14/c1-2-8-5-7-3-4-9(8)6-7/h2,7-9H,1,3-6H2
<b>InchiKey:</b>	PFUYSXSIHCSVJJ-UHFFFAOYSA-N
<b>Formula:</b>	C9H14
<b>SMILES:</b>	C=CC1CC2CCC1C2
<b>Mol. weight [g/mol]:</b>	122.21
<b>CAS:</b>	2146-39-6

## Physical Properties

Property code	Value	Unit	Source
gf	214.43	kJ/mol	Joback Method
hf	15.44	kJ/mol	Joback Method
hfus	13.03	kJ/mol	Joback Method
hvap	34.65	kJ/mol	Joback Method
log10ws	-2.51		Crippen Method
logp	2.609		Crippen Method
mcvol	111.650	ml/mol	McGowan Method
pc	3127.99	kPa	Joback Method
tb	415.08	K	Joback Method
tc	618.50	K	Joback Method
tf	217.55	K	Joback Method
vc	0.425	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	226.12	J/mol×K	415.08	Joback Method
cpg	244.15	J/mol×K	448.98	Joback Method
cpg	261.07	J/mol×K	482.89	Joback Method
cpg	276.92	J/mol×K	516.79	Joback Method
cpg	291.79	J/mol×K	550.69	Joback Method
cpg	305.71	J/mol×K	584.59	Joback Method
cpg	318.75	J/mol×K	618.50	Joback Method
dvisc	0.0005567	Paxs	217.55	Joback Method

dvisc	0.0005478	Paxs	250.47	Joback Method
dvisc	0.0005411	Paxs	283.39	Joback Method
dvisc	0.0005358	Paxs	316.32	Joback Method
dvisc	0.0005315	Paxs	349.24	Joback Method
dvisc	0.0005280	Paxs	382.16	Joback Method
dvisc	0.0005251	Paxs	415.08	Joback Method
hvapt	38.60	kJ/mol	367.50	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2146396&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2146396&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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