

# Benzene, 1-ethenyl-3-methoxy-

<b>Other names:</b>	3-Vinylanisole
<b>Inchi:</b>	InChI=1S/C9H10O/c1-3-8-5-4-6-9(7-8)10-2/h3-7H,1H2,2H3
<b>InchiKey:</b>	PECUPOXPPBBFLU-UHFFFAOYSA-N
<b>Formula:</b>	C9H10O
<b>SMILES:</b>	<chem>C=Cc1cccc(OC)c1</chem>
<b>Mol. weight [g/mol]:</b>	134.18
<b>CAS:</b>	626-20-0

## Physical Properties

Property code	Value	Unit	Source
gf	110.52	kJ/mol	Joback Method
hf	-10.82	kJ/mol	Joback Method
hfus	12.63	kJ/mol	Joback Method
hvap	40.31	kJ/mol	Joback Method
log10ws	-2.41		Crippen Method
logp	2.338		Crippen Method
mcvol	115.480	ml/mol	McGowan Method
pc	3299.15	kPa	Joback Method
tb	456.08	K	Joback Method
tc	667.91	K	Joback Method
tf	250.60	K	Joback Method
vc	0.430	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	288.32	J/molxK	667.91	Joback Method
cpg	278.87	J/molxK	632.60	Joback Method
cpg	268.85	J/molxK	597.30	Joback Method
cpg	258.24	J/molxK	561.99	Joback Method
cpg	247.03	J/molxK	526.69	Joback Method
cpg	235.20	J/molxK	491.38	Joback Method
cpg	222.73	J/molxK	456.08	Joback Method
dvisc	0.0016675	Paxs	250.60	Joback Method

dvisc	0.0001939	Paxs	456.08	Joback Method
dvisc	0.0002399	Paxs	421.83	Joback Method
dvisc	0.0003083	Paxs	387.59	Joback Method
dvisc	0.0004158	Paxs	353.34	Joback Method
dvisc	0.0005981	Paxs	319.09	Joback Method
dvisc	0.0009390	Paxs	284.85	Joback Method
hvapt	55.90	kJ/mol	393.50	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C626200&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C626200&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>h<sub>vapt</sub>:</b>	Enthalpy of vaporization at a given temperature
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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