

# 2-PROPYLCYCLOHEXANOL

|                      |                                                          |
|----------------------|----------------------------------------------------------|
| Other names:         | trans-2-Propylcyclohexanol                               |
| Inchi:               | InChI=1S/C9H18O/c1-2-5-8-6-3-4-7-9(8)10/h8-10H,2-7H2,1H3 |
| InchiKey:            | VZBNUCDUQJCIDP-UHFFFAOYSA-N                              |
| Formula:             | C9H18O                                                   |
| SMILES:              | CCCC1CCCCC1O                                             |
| Mol. weight [g/mol]: | 142.24                                                   |
| CAS:                 | 5846-43-5                                                |

## Physical Properties

| Property code | Value   | Unit    | Source             |
|---------------|---------|---------|--------------------|
| af            | 0.5212  |         | Cheméo Relay (1.0) |
| gf            | -95.18  | kJ/mol  | Joback Method      |
| hf            | -329.27 | kJ/mol  | Cheméo Relay (1.0) |
| hfus          | 16.06   | kJ/mol  | Joback Method      |
| hvap          | 70.42   | kJ/mol  | Cheméo Relay (1.0) |
| ie            | 9.56    | eV      | Cheméo Relay (1.0) |
| log10ws       | -2.42   |         | Cheméo Relay (1.0) |
| logp          | 2.338   |         | Crippen Method     |
| mcvol         | 132.680 | ml/mol  | McGowan Method     |
| pc            | 3025.61 | kPa     | Joback Method      |
| rinpol        | 1138.00 |         | NIST Webbook       |
| rinpol        | 1138.00 |         | NIST Webbook       |
| tb            | 478.23  | K       | Cheméo Relay (1.0) |
| tc            | 683.70  | K       | Cheméo Relay (1.0) |
| tf            | 256.75  | K       | Cheméo Relay (1.0) |
| vc            | 0.465   | m3/kmol | Cheméo Relay (1.0) |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 321.51 | J/mol×K | 512.38          | Joback Method |
| cpg           | 406.60 | J/mol×K | 700.47          | Joback Method |
| cpg           | 394.18 | J/mol×K | 669.12          | Joback Method |
| cpg           | 381.07 | J/mol×K | 637.77          | Joback Method |

|       |           |         |        |               |
|-------|-----------|---------|--------|---------------|
| cpg   | 367.27    | J/mol×K | 606.42 | Joback Method |
| cpg   | 352.75    | J/mol×K | 575.08 | Joback Method |
| cpg   | 337.50    | J/mol×K | 543.73 | Joback Method |
| dvisc | 0.0009760 | Pa×s    | 383.76 | Joback Method |
| dvisc | 0.0001611 | Pa×s    | 512.38 | Joback Method |
| dvisc | 0.0002632 | Pa×s    | 469.51 | Joback Method |
| dvisc | 0.0004745 | Pa×s    | 426.64 | Joback Method |
| dvisc | 0.0363477 | Pa×s    | 255.15 | Joback Method |
| dvisc | 0.0024066 | Pa×s    | 340.89 | Joback Method |
| dvisc | 0.0076936 | Pa×s    | 298.02 | Joback Method |

## Correlations

| Information                 | Value                         |
|-----------------------------|-------------------------------|
| Property code               | pvap                          |
| Equation                    | $\ln(P_{vp}) = A + B/(T + C)$ |
| Coeff. A                    | 1.38284e+01                   |
| Coeff. B                    | -3.82131e+03                  |
| Coeff. C                    | -6.80910e+01                  |
| Temperature range (K), min. | 350.30                        |
| Temperature range (K), max. | 516.76                        |

## Sources

|                                      |                                                                                                                                                                                         |
|--------------------------------------|-----------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|
| Crippen Method:                      | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                                                                       |
| Cheméo Relay (1.0):                  |                                                                                                                                                                                         |
| Cheméo Relay (1.0, beta):            |                                                                                                                                                                                         |
| The Yaws Handbook of Vapor Pressure: | <a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a> |
| NIST Webbook:                        | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C90676258&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C90676258&amp;Units=SI</a>                                           |
| Joback Method:                       | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                                                                   |
| McGowan Method:                      | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                                                                   |
| Crippen Method:                      | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>                                                                               |

## Legend

af: Acentric Factor

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>dvisc:</b>   | Dynamic viscosity                               |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>ie:</b>      | Ionization energy                               |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>pvap:</b>    | Vapor pressure                                  |
| <b>rinpol:</b>  | Non-polar retention indices                     |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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