

2-Propylcyclohexanol

Other names:	trans-2-Propylcyclohexanol
Inchi:	InChI=1S/C9H18O/c1-2-5-8-6-3-4-7-9(8)10/h8-10H,2-7H2,1H3
InchiKey:	VZBNUCDUQJCIDP-UHFFFAOYSA-N
Formula:	C9H18O
SMILES:	CCCC1CCCCC1O
Mol. weight [g/mol]:	142.24
CAS:	90676-25-8

Physical Properties

Property code	Value	Unit	Source
gf	-95.18	kJ/mol	Joback Method
hf	-347.34	kJ/mol	Joback Method
hfus	16.06	kJ/mol	Joback Method
hvap	52.43	kJ/mol	Joback Method
log10ws	-2.62		Crippen Method
logp	2.338		Crippen Method
mcvol	132.680	ml/mol	McGowan Method
pc	3025.61	kPa	Joback Method
rinpol	1138.00		NIST Webbook
tb	512.38	K	Joback Method
tc	700.47	K	Joback Method
tf	255.15	K	Joback Method
vc	0.490	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	321.51	J/mol×K	512.38	Joback Method
cpg	394.18	J/mol×K	669.12	Joback Method
cpg	381.07	J/mol×K	637.77	Joback Method
cpg	367.27	J/mol×K	606.42	Joback Method
cpg	352.75	J/mol×K	575.08	Joback Method
cpg	337.50	J/mol×K	543.73	Joback Method
cpg	406.60	J/mol×K	700.47	Joback Method

dvisc	0.0001611	Paxs	512.38	Joback Method
dvisc	0.0002632	Paxs	469.51	Joback Method
dvisc	0.0004745	Paxs	426.64	Joback Method
dvisc	0.0009760	Paxs	383.76	Joback Method
dvisc	0.0024066	Paxs	340.89	Joback Method
dvisc	0.0076936	Paxs	298.02	Joback Method
dvisc	0.0363477	Paxs	255.15	Joback Method

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.38284e+01
Coeff. B	-3.82131e+03
Coeff. C	-6.80910e+01
Temperature range (K), min.	350.30
Temperature range (K), max.	516.76

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C90676258&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions

h_{vap}:	Enthalpy of vaporization at standard conditions
log₁₀w_s:	Log10 of Water solubility in mol/l
log_p:	Octanol/Water partition coefficient
m_{cvol}:	McGowan's characteristic volume
p_c:	Critical Pressure
p_{vap}:	Vapor pressure
r_{inpol}:	Non-polar retention indices
t_b:	Normal Boiling Point Temperature
t_c:	Critical Temperature
t_f:	Normal melting (fusion) point
v_c:	Critical Volume

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