

# Oxirane, 2,2-dimethyl-3-(3,7,12,16,20-pentamethyl-3,7,11,15)

Other names:  
(all-E)-

2,2-Dimethyl-3-((3E,7E,11E,15E)-3,7,12,16,20-pentamethylhenicosa-3,7,11,15,19-pentae

InChI: InChI=1S/C30H50O/c1-24(2)14-11-17-27(5)20-12-18-25(3)15-9-10-16-26(4)19-13-21-28

InchiKey: QYIMSPSDBYKPPY-BANQPHDMSA-N

Formula: C30H50O

SMILES: CC(C)=CCCC(C)=CCCC(C)=CCCC=C(C)CCC=C(C)CCC1OC1(C)C

Mol. weight [g/mol]: 426.72

CAS: 7200-26-2

## Physical Properties

Property code	Value	Unit	Source
gf	521.50	kJ/mol	Joback Method
hf	-189.68	kJ/mol	Joback Method
hfus	68.80	kJ/mol	Joback Method
hvap	85.53	kJ/mol	Joback Method
log10ws	-10.85		Crippen Method
logp	9.816		Crippen Method
mcvol	407.070	ml/mol	McGowan Method
pc	738.42	kPa	Joback Method
rinpol	2963.40		NIST Webbook
rinpol	2963.40		NIST Webbook
tb	935.26	K	Joback Method
tc	1146.41	K	Joback Method
tf	396.83	K	Joback Method
vc	1.595	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1375.92	J/molxK	935.26	Joback Method
cpg	1403.12	J/molxK	970.45	Joback Method
cpg	1430.44	J/molxK	1005.64	Joback Method
cpg	1458.14	J/molxK	1040.84	Joback Method
cpg	1486.46	J/molxK	1076.03	Joback Method
cpg	1515.65	J/molxK	1111.22	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7200262&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7200262&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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