

(3S,8S,9S,10R,13R,14S,17R)-17-((2R,5R)-5-Ethyl-6

Inchi:	InChI=1S/C30H52O/c1-8-22(20(2)3)10-9-21(4)26-13-14-27-25-12-11-23-19-24(31-7)15-
InchiKey:	LOUBEXGURAKLSL-UHFFFAOYSA-N
Formula:	C30H52O
SMILES:	CCC(CCC(C)C1CCC2C3CC=C4CC(OC)CCC4(C)C3CCC12C)C(C)C
Mol. weight [g/mol]:	428.73
CAS:	20194-50-7

Physical Properties

Property code	Value	Unit	Source
gf	258.12	kJ/mol	Joback Method
hf	-534.42	kJ/mol	Joback Method
hfus	37.56	kJ/mol	Joback Method
hvap	81.86	kJ/mol	Joback Method
log10ws	-8.84		Crippen Method
logp	8.679		Crippen Method
mvol	391.690	ml/mol	McGowan Method
pc	848.01	kPa	Joback Method
rinpol	3276.90		NIST Webbook
rinpol	3276.90		NIST Webbook
tb	945.82	K	Joback Method
tc	1169.34	K	Joback Method
tf	507.61	K	Joback Method
vc	1.482	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1467.42	J/molxK	945.82	Joback Method
cpg	1500.33	J/molxK	983.07	Joback Method
cpg	1533.39	J/molxK	1020.33	Joback Method
cpg	1566.94	J/molxK	1057.58	Joback Method
cpg	1601.27	J/molxK	1094.83	Joback Method
cpg	1636.72	J/molxK	1132.09	Joback Method
cpg	1673.61	J/molxK	1169.34	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C20194507&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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