

# 5-Methyldocosane

<b>Other names:</b>	Docosane, 5-methyl
<b>Inchi:</b>	InChI=1S/C23H48/c1-4-6-8-9-10-11-12-13-14-15-16-17-18-19-20-22-23(3)21-7-5-2/h23H
<b>InchiKey:</b>	KDLUJNFDZQOZLK-UHFFFAOYSA-N
<b>Formula:</b>	C23H48
<b>SMILES:</b>	CCCCCCCCCCCCCCCCCCC(C)CCCC
<b>Mol. weight [g/mol]:</b>	324.63
<b>CAS:</b>	25163-52-4

## Physical Properties

Property code	Value	Unit	Source
gf	140.34	kJ/mol	Joback Method
hf	-523.33	kJ/mol	Joback Method
hfus	51.80	kJ/mol	Joback Method
hvap	66.40	kJ/mol	Joback Method
log10ws	-9.21		Crippen Method
logp	9.074		Crippen Method
mcvol	334.930	ml/mol	McGowan Method
pc	855.96	kPa	Joback Method
rinpol	2252.00		NIST Webbook
rinpol	2252.00		NIST Webbook
rinpol	2253.00		NIST Webbook
tb	725.20	K	Joback Method
tc	892.42	K	Joback Method
tf	333.97	K	Joback Method
vc	1.317	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1024.67	J/molxK	725.20	Joback Method
cpg	1047.11	J/molxK	753.07	Joback Method
cpg	1068.56	J/molxK	780.94	Joback Method
cpg	1089.04	J/molxK	808.81	Joback Method
cpg	1108.60	J/molxK	836.68	Joback Method

cpg	1127.28	J/mol×K	864.55	Joback Method
cpg	1145.09	J/mol×K	892.42	Joback Method
dvisc	0.0035428	Paxs	333.97	Joback Method
dvisc	0.0010359	Paxs	399.17	Joback Method
dvisc	0.0004278	Paxs	464.38	Joback Method
dvisc	0.0002197	Paxs	529.59	Joback Method
dvisc	0.0001306	Paxs	594.79	Joback Method
dvisc	0.0000860	Paxs	659.99	Joback Method
dvisc	0.0000611	Paxs	725.20	Joback Method
hvapt	75.60	kJ/mol	568.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.87881e+01
Coeff. B	-9.13136e+03
Coeff. C	1.20800e+00
Temperature range (K), min.	492.37
Temperature range (K), max.	676.36

## Sources

**The Yaws Handbook of Vapor Pressure:**  
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>  
<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C25163524&Units=SI>

## Legend

**cpg:** Ideal gas heat capacity  
**dvisc:** Dynamic viscosity  
**gf:** Standard Gibbs free energy of formation

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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