

# 7-Butyl-12-methylbenz[a]anthracene

|                      |                                                                                   |
|----------------------|-----------------------------------------------------------------------------------|
| Inchi:               | InChI=1S/C23H22/c1-3-4-10-21-20-13-8-7-11-18(20)16(2)23-19-12-6-5-9-17(19)14-15-2 |
| InchiKey:            | PBRXQWATBIWV-FV-UHFFFAOYSA-N                                                      |
| Formula:             | C23H22                                                                            |
| SMILES:              | CCCCc1c2ccccc2c(C)c2c1ccc1ccccc12                                                 |
| Mol. weight [g/mol]: | 298.42                                                                            |
| CAS:                 | 16354-51-1                                                                        |

## Physical Properties

| Property code | Value   | Unit    | Source         |
|---------------|---------|---------|----------------|
| gf            | 536.62  | kJ/mol  | Joback Method  |
| hf            | 245.81  | kJ/mol  | Joback Method  |
| hfus          | 38.87   | kJ/mol  | Joback Method  |
| hvap          | 76.64   | kJ/mol  | Joback Method  |
| log10ws       | -9.00   |         | Crippen Method |
| logp          | 6.797   |         | Crippen Method |
| mcvol         | 252.790 | ml/mol  | McGowan Method |
| pc            | 1707.53 | kPa     | Joback Method  |
| tb            | 829.18  | K       | Joback Method  |
| tc            | 1068.08 | K       | Joback Method  |
| tf            | 523.57  | K       | Joback Method  |
| vc            | 0.982   | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value     | Unit    | Temperature [K] | Source        |
|---------------|-----------|---------|-----------------|---------------|
| cpg           | 737.06    | J/mol×K | 829.18          | Joback Method |
| cpg           | 812.84    | J/mol×K | 1028.27         | Joback Method |
| cpg           | 798.77    | J/mol×K | 988.45          | Joback Method |
| cpg           | 784.32    | J/mol×K | 948.63          | Joback Method |
| cpg           | 769.33    | J/mol×K | 908.81          | Joback Method |
| cpg           | 753.63    | J/mol×K | 869.00          | Joback Method |
| cpg           | 826.68    | J/mol×K | 1068.08         | Joback Method |
| dvisc         | 0.0005210 | Pa×s    | 829.18          | Joback Method |
| dvisc         | 0.0005824 | Pa×s    | 778.25          | Joback Method |

|       |           |      |        |               |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0006613 | Pa×s | 727.31 | Joback Method |
| dvisc | 0.0007654 | Pa×s | 676.38 | Joback Method |
| dvisc | 0.0009072 | Pa×s | 625.44 | Joback Method |
| dvisc | 0.0011082 | Pa×s | 574.51 | Joback Method |
| dvisc | 0.0014075 | Pa×s | 523.57 | Joback Method |

## Sources

|                        |                                                                                                                                               |
|------------------------|-----------------------------------------------------------------------------------------------------------------------------------------------|
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C16354511&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C16354511&amp;Units=SI</a> |
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                     |
| <b>Crippen Method:</b> | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                             |
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                         |
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                         |

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>dvisc:</b>   | Dynamic viscosity                               |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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