

# 2,4-dichlorobenzyl hexadecyl ether

**Inchi:** InChI=1S/C23H38Cl2O/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-18-26-20-21-16-17-22(24)  
**InchiKey:** IVKTUGKMWWUJHF-UHFFFAOYSA-N  
**Formula:** C23H38Cl2O  
**SMILES:** CCCCCCCCCCCCCCOCC1CC(Cl)CC1Cl  
**Mol. weight [g/mol]:** 401.45

## Physical Properties

Property code	Value	Unit	Source
gf	107.07	kJ/mol	Joback Method
hf	-468.16	kJ/mol	Joback Method
hfus	58.17	kJ/mol	Joback Method
hvap	81.57	kJ/mol	Joback Method
log10ws	-9.51		Crippen Method
logp	8.991		Crippen Method
mcvol	341.520	ml/mol	McGowan Method
pc	966.87	kPa	Joback Method
rinpol	2800.00		NIST Webbook
rinpol	2800.00		NIST Webbook
rinpol	2800.00		NIST Webbook
rinpol	2808.00		NIST Webbook
rinpol	2813.00		NIST Webbook
rinpol	2800.00		NIST Webbook
rinpol	2804.00		NIST Webbook
rinpol	2808.00		NIST Webbook
rinpol	2804.00		NIST Webbook
rinpol	2812.00		NIST Webbook
rinpol	2806.00		NIST Webbook
rinpol	2800.00		NIST Webbook
rinpol	2813.00		NIST Webbook
rinpol	2800.00		NIST Webbook
rinpol	2810.00		NIST Webbook
tb	859.56	K	Joback Method
tc	1057.11	K	Joback Method
tf	482.50	K	Joback Method
vc	1.331	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1045.56	J/molxK	859.56	Joback Method
cpg	1063.93	J/molxK	892.49	Joback Method
cpg	1081.18	J/molxK	925.41	Joback Method
cpg	1097.36	J/molxK	958.34	Joback Method
cpg	1112.50	J/molxK	991.26	Joback Method
cpg	1126.66	J/molxK	1024.19	Joback Method
cpg	1139.88	J/molxK	1057.11	Joback Method
dvisc	0.0005303	Paxs	482.50	Joback Method
dvisc	0.0002671	Paxs	545.34	Joback Method
dvisc	0.0001550	Paxs	608.19	Joback Method
dvisc	0.0000996	Paxs	671.03	Joback Method
dvisc	0.0000690	Paxs	733.87	Joback Method
dvisc	0.0000507	Paxs	796.72	Joback Method
dvisc	0.0000390	Paxs	859.56	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R32290&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R32290&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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