

# 13H-Dibenz[bc,j]aceanthrylene

<b>Inchi:</b>	InChI=1S/C23H14/c1-2-7-18-14(4-1)10-11-19-20(18)12-17-9-8-15-5-3-6-16-13-21(19)23
<b>InchiKey:</b>	VHTSBQMSNJOJTG-UHFFFAOYSA-N
<b>Formula:</b>	C23H14
<b>SMILES:</b>	<chem>c1ccc2c(c1)ccc1c3c4c(ccc5cccc(c54)C3)cc12</chem>
<b>Mol. weight [g/mol]:</b>	290.36
<b>CAS:</b>	201-42-3

## Physical Properties

Property code	Value	Unit	Source
gf	726.30	kJ/mol	Joback Method
hf	530.87	kJ/mol	Joback Method
hfus	36.76	kJ/mol	Joback Method
hvap	78.82	kJ/mol	Joback Method
log10ws	-9.04		Crippen Method
logp	6.204		Crippen Method
mcvol	222.470	ml/mol	McGowan Method
pc	2318.07	kPa	Joback Method
tb	856.01	K	Joback Method
tc	1121.31	K	Joback Method
tf	598.01	K	Joback Method
vc	0.877	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	631.13	J/molxK	856.01	Joback Method
cpg	706.40	J/molxK	1077.10	Joback Method
cpg	690.07	J/molxK	1032.88	Joback Method
cpg	674.72	J/molxK	988.66	Joback Method
cpg	660.02	J/molxK	944.44	Joback Method
cpg	645.60	J/molxK	900.23	Joback Method
cpg	724.08	J/molxK	1121.31	Joback Method
dvisc	0.0055010	Paxs	856.01	Joback Method
dvisc	0.0056275	Paxs	813.01	Joback Method

dvisc	0.0057716	Paxs	770.01	Joback Method
dvisc	0.0059371	Paxs	727.01	Joback Method
dvisc	0.0061292	Paxs	684.01	Joback Method
dvisc	0.0063545	Paxs	641.01	Joback Method
dvisc	0.0066224	Paxs	598.01	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C201423&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C201423&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/73-729-2/13H-Dibenz-bc-j-aceanthrylene.pdf>

Generated by Cheméo on 2024-05-02 03:49:48.521537041 +0000 UTC m=+16911037.442114352.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.