

# Benzene, 1,1'-ethylidenebis[3,4-dimethyl-

**Other names:**

Ethane, 1,1-di-3,4-xylyl-  
1,1-Bis(3,4-dimethylphenyl)ethane  
1,1-Di-3,4-xylylethane  
Benzene, 1,1'-ethylidenebis\*3,4-dimethyl-  
«alpha»-(3,4-dimethylphenyl)-«alpha»-methyl-3,4-dimethyltoluene

**Inchi:** InChI=1S/C18H22/c1-12-6-8-17(10-14(12)3)16(5)18-9-7-13(2)15(4)11-18/h6-11,16H,1-5**InchiKey:** NCSVCMFDHINRJE-UHFFFAOYSA-N**Formula:** C18H22**SMILES:** Cc1ccc(C(C)c2ccc(C)c(C)c2)cc1C**Mol. weight [g/mol]:** 238.37**CAS:** 1742-14-9

## Physical Properties

Property code	Value	Unit	Source
gf	284.54	kJ/mol	Joback Method
hf	7.05	kJ/mol	Joback Method
hfus	25.38	kJ/mol	Joback Method
hvap	62.47	kJ/mol	Joback Method
log10ws	-5.86		Crippen Method
logp	5.072		Crippen Method
mcvol	216.960	ml/mol	McGowan Method
pc	1832.54	kPa	Joback Method
tb	684.08	K	Joback Method
tc	913.45	K	Joback Method
tf	380.54	K	Joback Method
vc	0.822	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	576.92	J/molxK	684.08	Joback Method
cpg	595.78	J/molxK	722.31	Joback Method
cpg	613.43	J/molxK	760.54	Joback Method
cpg	629.90	J/molxK	798.77	Joback Method

cpg	645.24	J/molxK	837.00	Joback Method
cpg	659.52	J/molxK	875.22	Joback Method
cpg	672.77	J/molxK	913.45	Joback Method
dvisc	0.0010382	Paxs	380.54	Joback Method
dvisc	0.0005770	Paxs	431.13	Joback Method
dvisc	0.0003628	Paxs	481.72	Joback Method
dvisc	0.0002491	Paxs	532.31	Joback Method
dvisc	0.0001826	Paxs	582.90	Joback Method
dvisc	0.0001407	Paxs	633.49	Joback Method
dvisc	0.0001126	Paxs	684.08	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	446.70	K	0.70	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1742149&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1742149&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume

<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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