

# 3,6,9,12,15,18,21,24-octaoxahexacosane-1,26-diol

<b>Other names:</b>	2-[2-[2-[2-[2-[2-[2-(2-Hydroxyethoxy)ethoxy]ethoxy]ethoxy]ethoxy]ethoxy]ethoxy]ethoxy]
<b>Inchi:</b>	InChI=1S/C18H38O10/c19-1-3-21-5-7-23-9-11-25-13-15-27-17-18-28-16-14-26-12-10-24
<b>InchiKey:</b>	YZUUTMGDONTGTN-UHFFFAOYSA-N
<b>Formula:</b>	C18H38O10
<b>SMILES:</b>	OCCOCCOCCOCCOCCOCCOCCOCCOCCOCCO
<b>Mol. weight [g/mol]:</b>	414.49
<b>CAS:</b>	3386-18-3

## Physical Properties

Property code	Value	Unit	Source
gf	-1012.96	kJ/mol	Joback Method
hf	-1777.07	kJ/mol	Joback Method
hfus	60.06	kJ/mol	Joback Method
hvap	108.30	kJ/mol	Joback Method
log10ws	1.42		Crippen Method
logp	-0.896		Crippen Method
mcvol	323.180	ml/mol	McGowan Method
pc	1190.70	kPa	Joback Method
rinpol	2936.50		NIST Webbook
rinpol	2936.50		NIST Webbook
tb	974.96	K	Joback Method
tc	1223.10	K	Joback Method
tf	592.10	K	Joback Method
vc	1.226	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1123.50	J/molxK	974.96	Joback Method
cpg	1139.18	J/molxK	1016.32	Joback Method
cpg	1152.16	J/molxK	1057.67	Joback Method
cpg	1162.37	J/molxK	1099.03	Joback Method
cpg	1169.71	J/molxK	1140.39	Joback Method
cpg	1174.08	J/molxK	1181.74	Joback Method

cpg	1175.39	J/molxK	1223.10	Joback Method
dvisc	0.0000149	Paxs	592.10	Joback Method
dvisc	0.0000052	Paxs	655.91	Joback Method
dvisc	0.0000022	Paxs	719.72	Joback Method
dvisc	0.0000011	Paxs	783.53	Joback Method
dvisc	0.0000006	Paxs	847.34	Joback Method
dvisc	0.0000003	Paxs	911.15	Joback Method
dvisc	0.0000002	Paxs	974.96	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3386183&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3386183&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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