

# 2,3,4,5,6-Pentamethylbenzophenone

|                      |                                                                                   |
|----------------------|-----------------------------------------------------------------------------------|
| Other names:         | 2,3,4,5,6-Pentamethyl benzopheneone                                               |
| Inchi:               | InChI=1S/C18H20O/c1-11-12(2)14(4)17(15(5)13(11)3)18(19)16-9-7-6-8-10-16/h6-10H,1- |
| InchiKey:            | JFFVBXWGARSRSC-UHFFFAOYSA-N                                                       |
| Formula:             | C18H20O                                                                           |
| SMILES:              | Cc1c(C)c(C)c(C(=O)c2ccccc2)c(C)c1C                                                |
| Mol. weight [g/mol]: | 252.35                                                                            |
| CAS:                 | 20386-33-8                                                                        |

## Physical Properties

| Property code | Value   | Unit    | Source         |
|---------------|---------|---------|----------------|
| gf            | 148.43  | kJ/mol  | Joback Method  |
| hf            | -111.72 | kJ/mol  | Joback Method  |
| hfus          | 30.11   | kJ/mol  | Joback Method  |
| hvap          | 70.27   | kJ/mol  | Joback Method  |
| log10ws       | -5.81   |         | Crippen Method |
| logp          | 4.460   |         | Crippen Method |
| mcvol         | 218.530 | ml/mol  | McGowan Method |
| pc            | 1896.95 | kPa     | Joback Method  |
| tb            | 743.37  | K       | Joback Method  |
| tc            | 974.83  | K       | Joback Method  |
| tf            | 457.99  | K       | Joback Method  |
| vc            | 0.834   | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value     | Unit    | Temperature [K] | Source        |
|---------------|-----------|---------|-----------------|---------------|
| cpg           | 594.97    | J/mol×K | 743.37          | Joback Method |
| cpg           | 611.49    | J/mol×K | 781.95          | Joback Method |
| cpg           | 626.87    | J/mol×K | 820.52          | Joback Method |
| cpg           | 641.16    | J/mol×K | 859.10          | Joback Method |
| cpg           | 654.40    | J/mol×K | 897.68          | Joback Method |
| cpg           | 666.63    | J/mol×K | 936.25          | Joback Method |
| cpg           | 677.89    | J/mol×K | 974.83          | Joback Method |
| dvisc         | 0.0007175 | Pa×s    | 457.99          | Joback Method |

|       |           |      |        |               |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0004697 | Pa×s | 505.55 | Joback Method |
| dvisc | 0.0003308 | Pa×s | 553.12 | Joback Method |
| dvisc | 0.0002462 | Pa×s | 600.68 | Joback Method |
| dvisc | 0.0001914 | Pa×s | 648.24 | Joback Method |
| dvisc | 0.0001540 | Pa×s | 695.81 | Joback Method |
| dvisc | 0.0001274 | Pa×s | 743.37 | Joback Method |

## Sources

|                        |                                                                                                                                               |
|------------------------|-----------------------------------------------------------------------------------------------------------------------------------------------|
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                         |
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                         |
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C20386338&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C20386338&amp;Units=SI</a> |
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                     |
| <b>Crippen Method:</b> | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                             |

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>dvisc:</b>   | Dynamic viscosity                               |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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