

# Nonadecane, 1-bromo-

<b>Other names:</b>	1-Bromononadecane Nonadecyl bromide
<b>Inchi:</b>	InChI=1S/C19H39Br/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-19-20/h2-19H2,1H1
<b>InchiKey:</b>	GWESGLGUMMNXDU-UHFFFAOYSA-N
<b>Formula:</b>	C19H39Br
<b>SMILES:</b>	CCCCCCCCCCCCCCCCCCCCBr
<b>Mol. weight [g/mol]:</b>	347.42
<b>CAS:</b>	4434-66-6

## Physical Properties

Property code	Value	Unit	Source
gf	123.42	kJ/mol	Joback Method
hf	-409.16	kJ/mol	Joback Method
hfus	50.25	kJ/mol	Joback Method
hvap	64.32	kJ/mol	Joback Method
log10ws	-8.21		Crippen Method
logp	8.033		Crippen Method
mvol	296.070	ml/mol	McGowan Method
pc	1131.38	kPa	Joback Method
tb	700.28	K	Joback Method
tc	871.42	K	Joback Method
tf	310.90 ± 4.00	K	NIST Webbook
vc	1.161	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	947.08	J/mol×K	871.42	Joback Method
cpg	931.61	J/mol×K	842.90	Joback Method
cpg	915.40	J/mol×K	814.38	Joback Method
cpg	898.43	J/mol×K	785.85	Joback Method
cpg	880.65	J/mol×K	757.33	Joback Method
cpg	862.03	J/mol×K	728.80	Joback Method
cpg	842.54	J/mol×K	700.28	Joback Method

dvisc	0.0021664	Paxs	363.69	Joback Method
dvisc	0.0000927	Paxs	700.28	Joback Method
dvisc	0.0001247	Paxs	644.18	Joback Method
dvisc	0.0001776	Paxs	588.08	Joback Method
dvisc	0.0002723	Paxs	531.99	Joback Method
dvisc	0.0004618	Paxs	475.89	Joback Method
dvisc	0.0009021	Paxs	419.79	Joback Method
hvapt	77.90	kJ/mol	583.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.51999e+01
Coeff. B	-5.61689e+03
Coeff. C	-1.21058e+02
Temperature range (K), min.	497.72
Temperature range (K), max.	689.09

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4434666&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4434666&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions

<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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