

Propionic acid, 3-iodo-, hexadecyl ester

| | |
|-----------------------------|---|
| Inchi: | InChI=1S/C19H37IO2/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-18-22-19(21)16-17-20/h2-1 |
| InchiKey: | AMBMHTVKDPBYIM-UHFFFAOYSA-N |
| Formula: | C19H37IO2 |
| SMILES: | CCCCCCCCCCCCCCCCOC(=O)CCI |
| Mol. weight [g/mol]: | 424.40 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | -66.70 | kJ/mol | Joback Method |
| hf | -603.42 | kJ/mol | Joback Method |
| hfus | 52.16 | kJ/mol | Joback Method |
| hvap | 76.42 | kJ/mol | Joback Method |
| log10ws | -7.59 | | Crippen Method |
| logp | 6.836 | | Crippen Method |
| mvol | 311.830 | ml/mol | McGowan Method |
| pc | 1097.90 | kPa | Joback Method |
| rinpol | 2578.00 | | NIST Webbook |
| rinpol | 2578.00 | | NIST Webbook |
| tb | 803.55 | K | Joback Method |
| tc | 992.65 | K | Joback Method |
| tf | 434.11 | K | Joback Method |
| vc | 1.212 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 919.94 | J/molxK | 803.55 | Joback Method |
| cpg | 999.85 | J/molxK | 961.14 | Joback Method |
| cpg | 985.65 | J/molxK | 929.62 | Joback Method |
| cpg | 970.59 | J/molxK | 898.10 | Joback Method |
| cpg | 954.64 | J/molxK | 866.58 | Joback Method |
| cpg | 937.77 | J/molxK | 835.07 | Joback Method |
| cpg | 1013.24 | J/molxK | 992.65 | Joback Method |
| dvisc | 0.0000613 | Paxs | 803.55 | Joback Method |

| | | | | |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0000820 | Paxs | 741.98 | Joback Method |
| dvisc | 0.0001158 | Paxs | 680.40 | Joback Method |
| dvisc | 0.0001749 | Paxs | 618.83 | Joback Method |
| dvisc | 0.0002896 | Paxs | 557.26 | Joback Method |
| dvisc | 0.0005433 | Paxs | 495.68 | Joback Method |
| dvisc | 0.0012187 | Paxs | 434.11 | Joback Method |

Sources

| | |
|------------------------|---|
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=U406247&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

Latest version available from:

<https://www.chemeo.com/cid/73-890-3/Propionic-acid-3-iodo-hexadecyl-ester.pdf>

Generated by Cheméo on 2024-05-04 06:06:14.103219087 +0000 UTC m=+17092023.023796411.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.