

# Fluoranthene, 1,2,3,10b-tetrahydro-

<b>Other names:</b>	1,2,3,4-Tetrahydrofluoranthene 1,2,3,10b-Tetrahydrofluoranthene
<b>Inchi:</b>	InChI=1S/C16H14/c1-2-8-13-12(7-1)14-9-3-5-11-6-4-10-15(13)16(11)14/h1-3,5,7-9,15H,
<b>InchiKey:</b>	VMBZUGDYWLFLEU-UHFFFAOYSA-N
<b>Formula:</b>	C16H14
<b>SMILES:</b>	<chem>c1ccc2c(c1)-c1cccc3c1C2CCC3</chem>
<b>Mol. weight [g/mol]:</b>	206.28
<b>CAS:</b>	20279-21-4

## Physical Properties

Property code	Value	Unit	Source
gf	433.18	kJ/mol	Joback Method
hf	243.34	kJ/mol	Joback Method
hfus	23.51	kJ/mol	Joback Method
hvap	57.54	kJ/mol	Joback Method
log10ws	-5.50		Crippen Method
logp	4.135		Crippen Method
mvol	167.060	ml/mol	McGowan Method
pc	2787.66	kPa	Joback Method
rinpol	1972.00		NIST Webbook
rinpol	314.60		NIST Webbook
rinpol	316.37		NIST Webbook
rinpol	328.21		NIST Webbook
rinpol	314.60		NIST Webbook
rinpol	1925.00		NIST Webbook
tb	643.39	K	Joback Method
tc	893.80	K	Joback Method
tf	407.64	K	Joback Method
vc	0.646	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	435.92	J/molxK	643.39	Joback Method

cpg	452.53	J/mol×K	685.13	Joback Method
cpg	467.79	J/mol×K	726.86	Joback Method
cpg	481.89	J/mol×K	768.60	Joback Method
cpg	495.03	J/mol×K	810.33	Joback Method
cpg	507.39	J/mol×K	852.07	Joback Method
cpg	519.18	J/mol×K	893.80	Joback Method
dvisc	0.0022525	Paxs	407.64	Joback Method
dvisc	0.0020138	Paxs	446.93	Joback Method
dvisc	0.0018333	Paxs	486.22	Joback Method
dvisc	0.0016926	Paxs	525.52	Joback Method
dvisc	0.0015801	Paxs	564.81	Joback Method
dvisc	0.0014884	Paxs	604.10	Joback Method
dvisc	0.0014123	Paxs	643.39	Joback Method
hvapt	68.00	kJ/mol	434.50	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C20279214&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C20279214&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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