

1,3-Dimethylanthracene

Inchi:	InChI=1S/C16H14/c1-11-7-12(2)16-10-14-6-4-3-5-13(14)9-15(16)8-11/h3-10H,1-2H3
InchiKey:	YPGZWANKCCQIBY-UHFFFAOYSA-N
Formula:	C16H14
SMILES:	Cc1cc(C)c2cc3ccccc3cc2c1
Mol. weight [g/mol]:	206.28
CAS:	610-46-8

Physical Properties

Property code	Value	Unit	Source
gf	380.66	kJ/mol	Joback Method
hf	210.69	kJ/mol	Joback Method
hfus	24.11	kJ/mol	Joback Method
hvap	58.75	kJ/mol	Joback Method
log10ws	-6.03		Crippen Method
logp	4.610		Crippen Method
mcvol	173.620	ml/mol	McGowan Method
pc	2555.92	kPa	Joback Method
rinpol	341.25		NIST Webbook
tb	645.06	K	Joback Method
tc	889.67	K	Joback Method
tf	399.46	K	Joback Method
vc	0.667	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	429.69	J/mol×K	645.06	Joback Method
cpg	497.31	J/mol×K	848.90	Joback Method
cpg	485.63	J/mol×K	808.13	Joback Method
cpg	473.15	J/mol×K	767.36	Joback Method
cpg	459.74	J/mol×K	726.60	Joback Method
cpg	445.30	J/mol×K	685.83	Joback Method
cpg	508.31	J/mol×K	889.67	Joback Method
dvisc	0.0004320	Pa×s	645.06	Joback Method

dvisc	0.0004850	Pa×s	604.13	Joback Method
dvisc	0.0005537	Pa×s	563.19	Joback Method
dvisc	0.0006454	Pa×s	522.26	Joback Method
dvisc	0.0007722	Pa×s	481.33	Joback Method
dvisc	0.0009553	Pa×s	440.39	Joback Method
dvisc	0.0012344	Pa×s	399.46	Joback Method

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.29338e+01
Coeff. B	-4.38417e+03
Coeff. C	-1.21066e+02
Temperature range (K), min.	467.74
Temperature range (K), max.	696.24

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C610468&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions

log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/74-003-6/1-3-Dimethylantracene.pdf>

Generated by Cheméo on 2025-12-20 20:45:56.75695256 +0000 UTC m=+6011754.286993216.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.