

# 1-(8-Bromo-2,3,6,7-tetrahydrodibenzo[1,2-b; 4,5-b']difuran-4-yl)-2-aminopropane, N-methyl,

**TFE**  
InchiKey:

InChI=1S/C16H17BrF3NO3/c1-8(21(2)15(22)16(18,19)20)7-11-9-3-5-24-14(9)12(17)10-4

Formula:

C16H17BrF3NO3

SMILES:

CC(Cc1c2c(c(Br)c3c1OCC3)OCC2)N(C)C(=O)C(F)(F)F

Mol. weight [g/mol]:

408.21

## Physical Properties

Property code	Value	Unit	Source
gf	-475.07	kJ/mol	Joback Method
hf	-893.19	kJ/mol	Joback Method
hfus	47.59	kJ/mol	Joback Method
hvap	77.35	kJ/mol	Joback Method
log10ws	-5.03		Crippen Method
logp	3.271		Crippen Method
mcvol	236.920	ml/mol	McGowan Method
pc	2127.56	kPa	Joback Method
rinpol	2380.00		NIST Webbook
rinpol	2336.00		NIST Webbook
rinpol	2336.00		NIST Webbook
tb	820.39	K	Joback Method
tc	1039.19	K	Joback Method
tf	587.99	K	Joback Method
vc	0.904	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	698.85	J/molxK	820.39	Joback Method
cpg	711.42	J/molxK	856.86	Joback Method
cpg	723.32	J/molxK	893.32	Joback Method
cpg	734.69	J/molxK	929.79	Joback Method
cpg	745.67	J/molxK	966.26	Joback Method
cpg	756.39	J/molxK	1002.73	Joback Method
cpg	766.98	J/molxK	1039.19	Joback Method

# Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R640505&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R640505&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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