

# Cyclopentane, undecyl-

<b>Other names:</b>	1-Undecylcyclopentane Undecane, 1-cyclopentyl- Undecylcyclopentane n-Undecylcyclopentane
<b>Inchi:</b>	InChI=1S/C16H32/c1-2-3-4-5-6-7-8-9-10-13-16-14-11-12-15-16/h16H,2-15H2,1H3
<b>InchiKey:</b>	FIWKORYTHDYMGF-UHFFFAOYSA-N
<b>Formula:</b>	C16H32
<b>SMILES:</b>	CCCCCCCCCCC1CCCC1
<b>Mol. weight [g/mol]:</b>	224.43
<b>CAS:</b>	6785-23-5

## Physical Properties

Property code	Value	Unit	Source
chl	-10557.10 ± 2.30	kJ/mol	NIST Webbook
gf	120.39	kJ/mol	Joback Method
hf	-312.50 ± 2.50	kJ/mol	NIST Webbook
hfus	31.13	kJ/mol	Joback Method
hvap	80.60	kJ/mol	NIST Webbook
log10ws	-6.17		Crippen Method
logp	6.098		Crippen Method
mcvol	225.440	ml/mol	McGowan Method
pc	1509.33	kPa	Joback Method
rinpol	1656.00		NIST Webbook
rinpol	1636.40		NIST Webbook
rinpol	1636.40		NIST Webbook
tb	580.76	K	Joback Method
tc	758.98	K	Joback Method
tf	280.98	K	Joback Method
vc	0.873	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	603.65	J/mol×K	580.76	Joback Method

cpg	625.27	J/molxK	610.46	Joback Method
cpg	645.89	J/molxK	640.17	Joback Method
cpg	665.52	J/molxK	669.87	Joback Method
cpg	684.22	J/molxK	699.57	Joback Method
cpg	702.00	J/molxK	729.28	Joback Method
cpg	718.91	J/molxK	758.98	Joback Method
dvisc	0.0050474	Paxs	280.98	Joback Method
dvisc	0.0019946	Paxs	330.94	Joback Method
dvisc	0.0010056	Paxs	380.91	Joback Method
dvisc	0.0005943	Paxs	430.87	Joback Method
dvisc	0.0003918	Paxs	480.83	Joback Method
dvisc	0.0002793	Paxs	530.80	Joback Method
dvisc	0.0002111	Paxs	580.76	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.61069e+01
Coeff. B	-5.22422e+03
Coeff. C	-9.92650e+01
Temperature range (K), min.	429.51
Temperature range (K), max.	583.19

## Sources

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C6785235&Units=SI>

**The Yaws Handbook of Vapor**

**Pressure:**

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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